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1 **Effect of steaming and *sous vide* processing on the total phenolic content,**
2 **vitamin C and antioxidant potential of the genus *Brassica***

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15 **Abbreviations**

16 ANOVA: Analysis of variance; DPPH: 2,2-Diphenyl-1-picrylhydrazyl; FRAP: Ferric reducing
17 antioxidant power; h^0 : Hue angle; HPLC: High-performance liquid chromatography; S.D:
18 Standard deviation; TPTZ: 2,4,6-Tris(2-pyridyl)-s-triazine; UV: Ultraviolet; TCEP: tris(2-
19 carboxyethyl)phosphine hydrochloride

20 **Abstract**

21 This study evaluated the effect of thermal processing on the antioxidant potential, vitamin
22 C, and total phenolic content of various parts, including edible co-products, of several
23 *Brassica* vegetables. Overall, no significant differences were observed in the lightness of
24 the samples after thermal processing, although the greenness and h^0 values of the samples
25 were affected ($p<0.05$). Similar profiles were observed for the leaves, inflorescences, and
26 stalks of the studied crucifers. The stalks of some varieties, including broccoli cv.
27 Parthenon and kale cv. Crispa, showed higher vitamin C contents compared to that of
28 their inflorescences ($p<0.05$). Both steaming and *sous-vide* processing significantly
29 reduced the vitamin C and total phenolic content of the crucifers studied ($p<0.05$). The
30 results demonstrated that *Brassica* co-products contain valuable and health-promoting
31 substances that can be lost during thermal processing; this must be considered when
32 calculating the dietary intake of these compounds from cooked vegetables.

33 **Industrial relevance**

34 The results obtained herein suggest that *Brassica* stalks are as nutritious and healthy as
35 the florets or leaves, which are more commonly consumed as part of our diet. In addition,
36 this study demonstrates the potential of *Brassica* co-products as a resource for the
37 extraction of antioxidant compounds and opens new commercial opportunities for their
38 use beyond their current applications in the food industry. The results also highlight that
39 these compounds are mostly lost during processing and that the processing conditions
40 should be carefully optimized to minimize their degradation.

41 **Keywords:** thermal processing, *sous vide*, antioxidant activity, vitamin C, ascorbic acid, phenolic
42 content, food co-products, cruciferous vegetables, waste reduction

43 1. Introduction

44 To aid processing, extend shelf-life and to obtain low-cost and good-tasting foods, the
45 food industry has relied on the use of fats, sugars, chemical processing aids, and plastics
46 (Lamppa, Horn, & Edwards, 2014). However, consumers are now becoming more aware
47 of the relationship among food, diet, and health, and this has led to increased interest in
48 natural ingredients and the consumption of food products that are tasty, nutritious and
49 healthy (Grasso, Brunton, Lyng, Lalor, & Monahan, 2014). The results obtained from
50 several epidemiological studies have encouraged a high intake of plant products, which
51 are associated with a reduced risk for several chronic diseases, such as atherosclerosis
52 and cancer (Podsędek, 2007). Furthermore, the number of consumers following a vegan
53 diet and the demand for vegan food have notably increased in many countries, and it is
54 likely that this trend will continue to grow (Janssen, Busch, Rödiger, & Hamm, 2016).

55 The family *Brassicaceae* (or *Cruciferae*) consists of 350 genera and over 3500 species,
56 which include the genera *Camelina*, *Crambe*, *Sinapis*, and *Brassica* (Cartea, Francisco,
57 Soengas, & Velasco, 2010). *Brassica* plants are informally known as cruciferous
58 vegetables, crucifers, cabbages, or mustard plants and they are economically the most
59 important genus in the *Brassicaceae* family (Rakow, 2004). *Brassica* species that are
60 commonly used for food include broccoli, cauliflower, cabbage (*B. oleracea*), and turnip
61 (*B. rapa*). These vegetables are known for their anti-carcinogenic properties (Podsędek,
62 2007; Singh, Upadhyay, Prasad, Bahadur, & Rai, 2007), and several studies recently
63 reported the occurrence of antioxidant compounds in several crucifers, such as Chinese
64 cabbage (Seong, Hwang, & Chung, 2016) and red and green cabbages (Upadhyay,
65 Sehwaq, & Singh, 2016).

66 Research on cruciferous vegetables has focused mainly on the plant parts that are most
67 commonly consumed. The non-consumed parts, which are the crucifer co-products, are

68 usually discarded as waste or used for low-value purposes (Ngu & Ledin, 2005). It is not
69 practical to discard co-products and wastes, especially when these contain a significant
70 amount of bioactive compounds that can promote human health and a nutritious diet.
71 Waste revalorization and the reutilization of co-products are important issues in the food
72 industry (Lafarga, Gallagher, Walsh, Valverde, & Hayes, 2013). Furthermore, although
73 some crucifers can be eaten fresh, these vegetables are most commonly consumed after
74 cooking. The temperature may affect the antioxidant content of foods due to antioxidant
75 release, destruction, or even the creation of new metabolites (Wachtel-Galor, Wong, &
76 Benzie, 2008). There is a need for quantitative data on the nutritional and bioactive
77 properties of cooked *Brassica* vegetables. The effects of cooking and cooking methods
78 on the bioactive properties of vegetables have not been well studied. Such information
79 would not only help to better understand the function of antioxidant phytochemicals but
80 also promote their consumption and promote health.

81 In this work, the authors report the effect of two cooking methods (a conventional thermal
82 treatment and *sous-vide*) on the colour, antioxidant activity, total phenolic content, and
83 total vitamin C content of different *Brassica* vegetables, including broccoli, cauliflower,
84 and cabbage. The analysed parts of the plants included those that are regularly consumed
85 and those that are not currently used as food sources, opening new commercial
86 opportunities for their use in the food industry.

87 **2. Materials and methods**

88 **2.1 Chemicals and reagents**

89 Methanol, sodium acetate, acetic acid, hydrochloric acid, and ferric chloride hexahydrate
90 were obtained from Panreac (Barcelona, Spain). Gallic acid, ascorbic acid,
91 metaphosphoric acid, 2,4,6-tris(2-pyridyl)-s-triazine (TPTZ), 2,2-diphenyl-1-
92 picrylhydrazyl (DPPH), tris(2-carboxyethyl)phosphine hydrochloride (TCEP), and
93 sodium carbonate were purchased from Sigma-Aldrich (Steinheim, Germany). Folin-
94 Ciocalteu's reagent was purchased from VWR (Llinars del Vallès, Spain). All reagents
95 used were of analytical grade.

96 **2.2 Plant material: Collection and processing**

97 *Brassica* vegetables, including broccoli cv. Marathon (*Brassica oleracea* var. *italica*),
98 broccoli cv. Parthenon (*Brassica oleracea* var. *italica*), broccoli cv. Graffiti (*Brassica*
99 *oleracea* var. *botrytis*), broccoli cv. Pastoret (*Brassica oleracea* var. *botrytis*), Espigall
100 del Garraf (*Brassica oleracea* var. *acephala*), and kale cv. Crispa (*Brassica oleracea* var.
101 *acephala*), were provided by the Fundació Miquel Agustí, Barcelona, Spain. The plants
102 were grown at Agròpolis, Baix Llobregat, Barcelona, Spain (41°17'18.6"N 2°02'39.7"E)
103 and collected during November 2015.

104 The sample processing was performed at the pilot plant of the IRTA Fruitcentre in Lleida,
105 Spain. Upon collection, the samples were frozen using liquid nitrogen and stored at -80
106 °C until further use. Sample processing consisted of dividing the samples into six
107 replicates of approximately 100 g each of either the leaves/stems or inflorescences/stems.
108 Three replicates were used for the steaming and *sous-vide* treatments.

109 Before the *sous-vide* treatment, the samples were rinsed using tap water for 10 seconds
110 and vacuum-sealed in a polyethylene vacuum-sealable bags designed for this treatment.
111 The samples were vacuum-sealed using a “soft vacuum” programme. Both treatments,
112 the conventional processing and *sous-vide*, were performed in a Rational SCC WE-101
113 convection oven (Rational AG, Landsberg am Lech, Germany). The treatment conditions
114 for steaming were: 100 °C and 15 min for inflorescences or leaves and 100 °C and 65 min
115 for stems. The treatment conditions for the *sous vide*-treated samples were: 80 °C and 15
116 min for inflorescences or leaves and 80 °C and 90 min for stems. After treatment, the
117 samples were quickly chilled to approximately 3-4 °C before freezing using liquid
118 nitrogen and storage at -80 °C.

119 **2.3 Colour measurement**

120 Eight colour recordings were taken per part of the plant (stem, leaf, or inflorescence) and
121 per treatment (fresh and after the conventional treatment or *sous-vide*) for each sample
122 using a Minolta CR-200 colorimeter (Minolta INC, Tokyo, Japan). The CIE values were
123 recorded in terms of L^* (lightness), a^* (redness, greenness), and b^*
124 (yellowness/blueness). A calibration was performed using a standard white tile (Y:92.5,
125 x:0.3161, y:0.3321) provided by the manufacturer and the D65 illuminant, which
126 approximates daylight.

127 The hue angle (h^0) was calculated with the obtained L^* , a^* , and b^* values using the
128 equation:

$$129 \quad h^0 = \tan^{-1} \frac{b^*}{a^*}$$

130 **2.4 Determination of the content of vitamin C**

131 The total vitamin C content (ascorbic acid and dehydroascorbic acid) was determined in
132 triplicate with high-performance liquid chromatography (HPLC) using a Waters 717 plus
133 Autosampler HPLC system (Waters Corp., NJ, USA) coupled to an ultraviolet (UV)
134 detector following the method previously described by Plaza et al. (2011). Briefly, six
135 grams of a frozen sample were homogenized with 20 mL of an extraction solution that
136 contained 30 g/L meta-phosphoric acid and 80 mL/L acetic acid in HPLC-grade water.
137 The resulting mixture was centrifuged using a Sigma 3-18KS centrifuge (Osterode am
138 Harz, Germany) at 12,000 rpm for 20 min at 4 °C and filtered, and the total volume was
139 adjusted to 25 mL with the extraction solution. The samples were further filtered through
140 0.45- μ m filters, and an aliquot (1.9 mL) of the mixture was obtained to react with 0.1 mL
141 of 0.04 M TCEP for 3 h at room temperature in the dark.

142 Vitamin C was separated on a reversed-phase SupelcosilTM LC18 (5 μ m) stainless-steel
143 column (250 \times 4.6 mm i.d., Supelco, USA). An isocratic solvent system was used (0.1
144 mL/L of sulphuric acid, pH 2.5-2.6). The flow rate was fixed at 1 mL/min, and the UV-
145 vis photodiode array detector was set at 254 nm. The identification of vitamin C was
146 performed by comparing the retention time and the obtained spectra to those previously
147 obtained with a standard.

148 **2.5 Determination of the total phenolic content**

149 The total phenolic content was determined using Folin-Ciocalteu's method following the
150 modifications described by Altisent, Plaza, Alegre, Viñas, and Abadias (2014). Briefly,
151 for the extraction, six grams of a frozen, fresh sample were homogenized with 20 mL of
152 methanol 70% (v/v), centrifuged using a Sigma 3-18KS centrifuge (Osterode am Harz,
153 Germany) at 12,000 rpm for 20 min at 4 °C, and filtered. The extraction solution was
154 added to the extract to obtain a final volume of 25.0 mL. The assay was performed in
155 triplicate by adding 4.3 mL of deionized water and 0.5 mL of Folin-Ciocalteu's reagent

156 to 0.7 mL of each extract. After 5 min of incubation, 2.0 mL of a saturated sodium
157 carbonate solution was added. The mixture was shaken and further incubated for 1 h in
158 the dark, and the absorbance was read at 760 nm using a GENESYS™ 10S UV-Vis
159 spectrophotometer (Thermo Fisher Scientific, MA, USA). The results are expressed as
160 mg of gallic acid per kilogram of fresh weight.

161 **2.6 Antioxidant activity**

162 The antioxidant activity was measured using two different methods: the ferric reducing
163 antioxidant power (FRAP) and DPPH· scavenging activity assay. The extraction
164 methodology was the same for both methods. A sample extract was obtained by mixing
165 20 mL of methanol 70% (v/v) with six grams of sample and followed by homogenization
166 for 5 min and centrifugation using a Sigma 3-18KS centrifuge (Osterode am Harz,
167 Germany) at 12,000 rpm for 20 min at 4 °C. The extracts were diluted to 25 mL with the
168 extraction solution.

169 **2.6.1 Ferric reducing antioxidant powder (FRAP)**

170 The total antioxidant potential of the samples was determined for each sample using the
171 FRAP assay previously described by Benzie and Strain (1996). Briefly, the FRAP reagent
172 was freshly prepared by mixing 0.3 M acetate buffer (pH 3.6), 10 mM TPTZ in 40 mM
173 HCl and 20 mM FeCl₃ in the proportion 10:1:1 (v/v/v), respectively. The assay was
174 performed by adding 1.4 mL of the FRAP reagent to 0.1 mL of the generated extract.
175 After 20 min of incubation in the dark at 37 °C with shaking, the absorbance was read at
176 593 nm using a GENESYS™ 10S UV-Vis spectrophotometer (Thermo Fisher Scientific,
177 MA, USA). The results were run in triplicate and compared to a standard curve prepared
178 daily with different concentrations of ascorbic acid.

179 **2.6.2 The 2,2-diphenyl-1-picrylhydrazyl radical (DPPH·) scavenging activity**

180 The antioxidant activity was also measured following the method described by Hidalgo,
181 Sánchez-Moreno, and de Pascual-Teresa (2010). Briefly, the assay was performed by
182 adding 1.4 mL of a 0.1 mM DPPH solution to 0.1 mL of the generated extract. After 60
183 min of incubation at room temperature in the dark, the absorbance was read at 515 nm
184 using a GENESYS™ 10S-UV Vis spectrophotometer (Thermo Fisher Scientific, MA,
185 USA). The results were run in triplicate and compared to a standard curve prepared daily
186 with different concentrations of ascorbic acid.

187 **2.7 Statistical analysis**

188 All tests were replicated three times, except for the colour readings, which were recorded
189 eight times per sample. The results are expressed as the mean \pm standard deviation (S.D.).
190 Samples were analysed using analysis of variance (ANOVA). Statistical analysis was
191 done using JMP 8 (SAS Institute Inc., Cary, USA). Tukey's pairwise comparison of the
192 means was conducted to identify differences between treatments, and Student's t-test was
193 used to identify differences between different parts of the same crucifer. The criterion for
194 statistical significance was $p < 0.05$.

195 3. Results and discussion

196 3.1 Effect of thermal processing on the colour

197 The present study evaluated the effect of thermal processing on the colour of selected
198 *Brassica* vegetables. Colour is a key parameter in food products because food colour is
199 important for humans' innate perceptions of the value of food items and is the first
200 parameter of quality evaluated by consumers (Markovic, Ilic, Markovic, Simonovic, &
201 Kosanic, 2013). Colour is a reliable indicator of the healthful quality of foods. In this
202 study, eight recordings of the L^* , a^* , and b^* values were taken per part per plant. L^* is
203 defined as the lightness ($L^*=0$ yields black, and $L^*=100$ indicates diffuse white).

204 Table 1 shows the L^* and h^0 values, which were calculated from the a^* and b^* values.
205 The colours of the florets and stems of the raw and thermally processed crucifers were
206 comparable to those obtained in previous studies (Brewer, Begum, & Bozeman, 1995;
207 Miglio, Chiavaro, Visconti, Fogliano, & Pellegrini, 2007). In the current study, no
208 significant differences were observed between the lightness of the inflorescences and the
209 stalks of the fresh broccoli species. Only broccoli cv. Pastoret presented a higher L^* value
210 in the stalk compared to the inflorescence ($p<0.05$). Thermal processing did not affect the
211 lightness of the inflorescences of the different broccoli varieties and the leaves of Espigall
212 del Garraf. However, steaming affected the lightness of the leaves of kale cv. Crispa
213 ($p<0.05$). In addition, thermal processing resulted in a significant increase in the L^* value
214 of the stalks of all the studied varieties ($p<0.05$). The results contrast with previous
215 studies, which reported a decrease in the lightness of broccoli after thermal processing
216 (Miglio, et al., 2007). Pellegrini et al. (2010) also reported differences in the lightness of
217 *Brassica* vegetables after thermal processing.

218 No differences were observed between the h^0 value of the fresh broccoli inflorescences
219 or leaves of Espigall del Garraf and their stalks, except for kale cv. Crispa, which showed
220 a higher h^0 value for the leaves ($p<0.05$). Overall, thermal processing significantly
221 reduced the h^0 value in both the inflorescences and stalks, except for the inflorescences
222 of broccoli cv. Graffiti and the stalks of broccoli cv. Parthenon and broccoli cv. Pastoret.
223 A decrease in the h^0 value is associated with a loss of greenness. Similar results were
224 obtained by Miglio et al. (2007), who observed a loss of greenness and a reduction in the
225 h^0 value of *Brassica* samples after both steaming and frying. During processing,
226 chlorophyll is converted into pheophytin and pyropheophytin, turning vegetables from a
227 bright green to an olive green colour (Bongoni, Verkerk, Steenbekkers, Dekker, &
228 Stieger, 2014). The observed reduction in greenness in the h^0 value of the crucifers after
229 thermal processing could be caused not only by the degradation of chlorophyll but also
230 by changes in the surface reflectance and depth of light penetration in thermally processed
231 vegetables, which are caused by a loss of air and other dissolved gases (Tijskens,
232 Schijvens, & Biekman, 2001). Although processing resulted in a reduction of the h^0 value
233 in most samples, *sous-vide* processing of the stalks of broccoli cv. Marathon and broccoli
234 cv. Graffiti resulted in higher h^0 values ($p<0.05$). This trend was also previously observed
235 after steaming crucifer stems (Miglio, et al., 2007).

236 **3.2 Effect of thermal processing on the vitamin C content**

237 Vitamin C includes ascorbic acid, and its oxidation product, dehydroascorbic acid, has
238 several biological activities in the human body and is thought to have cancer-protective
239 capacities (Bakker, et al., 2016). Over 85% of vitamin C in human diets is supplied by
240 fruits and vegetables (Podsędek, 2007). Indeed, the concentration of vitamin C in blood
241 is an excellent biomarker of vegetable and fruit consumption and provides better

242 approximations of the concentration available to cells than dietary questionnaires (Block,
243 Norkus, Hudes, Mandel, & Helzlsouer, 2001). However, the content of vitamin C in fruits
244 and vegetables can be significantly reduced during processing and storage due to its
245 solubility in water and its sensitivity to high temperatures and oxidation conditions
246 (Gamboa-Santos, Cristina Soria, Pérez-Mateos, Carrasco, Montilla, & Villamiel, 2013).
247 For example, conventional cooking of broccoli for 0.5, 1.5, and 5.0 min results in vitamin
248 C losses of 19.2, 47.5, and 65.9%, respectively (Zhang & Hamazu, 2004). Vitamin C
249 loss caused by food processing can be reduced. For example, Lin and Brewer (2005)
250 observed that steam blanching resulted in better vitamin C retention in peas compared to
251 treatments with boiling water for equal blanching times.

252 The vitamin C content of *Brassica* vegetables varies significantly between subspecies.
253 For example, Pfendt, Vukašinović, Blagojević, and Radojević (2003) reported the
254 ascorbic acid content of kale as 92.6 mg/100 g of edible portion. This value is much higher
255 than the ascorbic acid content of white cabbage reported by Bahorun, Luximon-Ramma,
256 Crozier, and Aruoma (2004), 8 mg/100 mg edible portion. In the current study, broccoli
257 cv. Graffiti florets showed the highest vitamin C content with a concentration of $220.3 \pm$
258 17.1 mg/100 g of fresh sample ($p < 0.05$). The vitamin C contents of the studied fresh and
259 processed vegetables are shown in Figure 1. Overall, the inflorescences/leaves of fresh
260 vegetables presented higher vitamin C contents than the stem. However, for some
261 varieties, including broccoli cv. Parthenon and kale cv. Crispa, the vitamin C contents of
262 the fresh stalks were higher ($p < 0.05$). The results compared favourably with those
263 obtained previously by Zhang et al. (2004) who reported a higher vitamin C content in
264 the stem than the floret of broccoli. Thermal processing significantly reduced the vitamin
265 C content of all analysed samples ($p < 0.05$). The results are in line with previous studies
266 that reported a reduction in the vitamin C content of vegetables after thermal processing

267 (Wachtel-Galor, et al., 2008). The observed reduction was especially high in the
268 inflorescences of broccoli cv. Marathon, broccoli cv. Graffiti, and broccoli cv. Pastoret
269 and in the leaves of Espigall del Garraf compared to the reduction in the vitamin C
270 contents of their stems ($p<0.05$). The vitamin C contents of the stems of broccoli cv.
271 Marathon, broccoli cv. Graffiti, broccoli cv. Pastoret, Espigall del Garraf, and kale cv.
272 Crispa were higher than those observed in their inflorescences/leaves after both steaming
273 and *sous-vide* processing ($p<0.05$). However, this trend was not observed in broccoli cv.
274 Parthenon, and no differences were observed between the two plant parts after processing.
275 Steaming resulted in higher vitamin C losses in both the stems and leaves of kale cv.
276 Crispa compared to *sous-vide* processing ($p<0.05$). This could be caused by the reduced
277 amount of oxygen present when cooking by *sous-vide*, and previous studies have
278 suggested that oxygen is probably the most important factor in vitamin C degradation
279 (Verbeyst, Bogaerts, Van der Plancken, Hendrickx, & Van Loey, 2013).

280 **3.3 Effect of processing on the total phenolic content**

281 Natural antioxidants can be divided into three main groups: vitamins, carotenoids, and
282 phenolic compounds (Thaipong, Boonprakob, Crosby, Cisneros-Zevallos, & Hawkins
283 Byrne, 2006). Polyphenols possess ideal structural chemistry for free radical-scavenging
284 activities. Over 8,000 phenolic compounds, including phenolic acids and flavonoids, have
285 been identified in various plant species and have been linked to anti-diabetic, anti-ageing,
286 anti-cancer, neuro-protective, and cardio-protective effects (Pandey & Rizvi, 2009).
287 Although several authors published extensive reviews on the phenolic profiles in different
288 *Brassica* species (Cartea, et al., 2010; Podsędek, 2007), data on the phenolic compounds
289 in plant parts that are not commonly consumed are scarce. For this reason, this study
290 aimed to quantify the phenolic compounds in the various parts of these vegetables. Figure
291 2 shows the total phenolic content of the studied *Brassica* vegetables. The highest

292 phenolic content was found in the fresh leaves of kale cv. Crispa, which had a total
293 phenolic content of 158.8 ± 3.5 mg/100 g sample. Results obtained herein were higher
294 when compared to those reported by Leja, Mareczek, Starzyńska, and Rożek (2001) who
295 observed that broccoli florets contained 56.2 mg TPC/100 g of fresh weight. Similar
296 results were also reported by Zhang et al. (2004) who published a concentration of 34.5
297 mg TPC/100 g of fresh broccoli. No differences were observed between the total phenolic
298 content of the stems and the inflorescences of fresh broccoli cv. Parthenon and broccoli
299 cv. Pastoret. The inflorescences and leaves of broccoli cv. Graffiti, Espigall del Garraf,
300 and kale cv. Crispa showed a higher total phenolic content than their stems ($p < 0.05$). This
301 trend was not observed in broccoli cv. Marathon, which showed a higher phenolic content
302 in its stem than the inflorescence ($p < 0.05$). Thermal processing significantly reduced the
303 phenolic content of the studied vegetables. Similar results were obtained previously (dos
304 Reis, de Oliveira, Hagen, Jablonski, Flôres, & de Oliveira Rios, 2015; Pellegrini, et al.,
305 2010). Other previously studied treatments, such as microwaving, also resulted in high
306 losses of flavonoids (97%), sinapic acid derivatives (74%), and caffeoyl-quinic acid
307 derivatives (87%) in *Brassica* species (Vallejo, Tomás-Barberán, & García-Viguera,
308 2003). Overall, no differences were observed between the samples treated by steaming or
309 *sous-vide* processing for most of the varieties. However, the phenolic contents of the
310 stems of broccoli cv. Parthenon and broccoli cv. Pastoret treated using *sous-vide* were
311 significantly higher than those obtained after steaming ($p < 0.05$). However, the opposite
312 trend was observed after processing the inflorescences of broccoli cv. Parthenon, and
313 *sous-vide* processing resulted in a higher phenolic content loss ($p < 0.05$).

314 **3.4 Measurement of the antioxidant activity**

315 Natural antioxidants in fruits and vegetables have gained increasing interest over the last
316 decade (Thaipong, et al., 2006). As mentioned previously, several studies have

317 highlighted the antioxidant potential of *Brassica* species (Bekhit, Lingming, Mason,
318 Zhou, & Sedcole, 2013; Podsędek, 2007; Seong, et al., 2016; Upadhyay, et al., 2016;
319 Wachtel-Galor, et al., 2008). Although some studies suggested that plant parts that are
320 not commonly consumed have similar antioxidant potentials to those that are eaten
321 (Balasundram, Sundram, & Samman, 2006; Wijngaard, Rößle, & Brunton, 2009), most
322 studies have focused on the antioxidant potential of commonly consumed parts. The
323 current study evaluated the antioxidant potential of several *Brassica* species using two
324 independent methods: the FRAP assay and DPPH radical assay.

325 The FRAP assay uses antioxidants as reductants in a redox-linked colorimetric method
326 employing an easily reduced oxidant. The ferric to ferrous ion reduction at low pH values
327 causes the formation of a coloured ferrous-tripyridyltriazine complex (Benzie, et al.,
328 1996). Figure 3 shows the antioxidant activity of the fresh and thermally treated samples
329 measured using the FRAP method. The inflorescences of fresh broccoli cv. Marathon and
330 broccoli cv. Graffiti and the leaves of Espigall del Garraf and kale cv. Crispa showed
331 higher antioxidant activities, as measured using the FRAP assay, compared to their stems
332 ($p<0.05$). However, the opposite trend was observed in the samples of broccoli cv.
333 Pastoret, which presented a higher antioxidant activity in the stems ($p<0.05$); no
334 differences were observed between the stems and inflorescences of broccoli cv.
335 Parthenon. *Sous-vide* processing resulted in a significant reduction in the antioxidant
336 potential of the inflorescences and leaves of all the studied vegetables ($p<0.05$). However,
337 no differences were observed between the antioxidant activities of fresh and *sous-vide*-
338 treated stems of broccoli cv. Parthenon, Espigall del Garraf, and kale cv. Crispa, and an
339 increase was observed in the antioxidant potential of the stems of broccoli cv. Marathon
340 after both steaming and *sous-vide* processing compared to that of the fresh stems
341 ($p<0.05$). For some parts of some varieties, including the inflorescences of broccoli cv.

342 Marathon and broccoli cv. Parthenon, *sous-vide* processing resulted in a higher loss of
343 antioxidant potential compared to steaming. However, for the stems of broccoli cv.
344 Parthenon and the leaves of Espigall del Garraf and kale cv. Crispa, a significantly higher
345 loss of antioxidant potential was observed after steaming ($p<0.05$).

346 The results obtained using the DPPH· assay, as shown in Figure 4, did not correlate well
347 to those obtained using the FRAP method in terms of the antioxidant potential. The
348 antioxidant potential was higher in the inflorescences of fresh broccoli cv. Graffiti and
349 kale cv. Crispa compared to their fresh stems ($p<0.05$), and no differences were observed
350 between the inflorescences or leaves and stems of fresh broccoli cv. Parthenon, broccoli
351 cv. Marathon, and Espigall del Garraf. Overall, the antioxidant activity of the analysed
352 samples measured using the DPPH· assay increased after thermal processing. This
353 increase in antioxidant activity was higher after the *sous-vide* processing of the
354 inflorescences of broccoli cv. Marathon and the stems and inflorescences of broccoli cv.
355 Pastoret and after steaming the inflorescences of broccoli cv. Pastoret and the stems of
356 broccoli cv. Parthenon and Espigall del Garraf. These results are in line with those
357 obtained previously by Juárez et al. (2016) and by Wachtel-Galor et al. (2008) who
358 reported an increase in the antioxidant activity after cooking several *Brassica* vegetables.
359 Similar results were also obtained by Turkmen, Sari, and Velioglu (2005) who observed
360 an increase in the antioxidant activity of broccoli after boiling, microwaving, and
361 steaming. The observed increase in the antioxidant activity could be caused by the
362 liberation of antioxidants from insoluble portions or the formation of novel antioxidants
363 caused by temperature-dependent reactions (Hwang, Shin, Lee, Lee, & Yoo, 2012;
364 Manzocco, Calligaris, Mastrocola, Nicoli, & Lerici, 2000; Martins, Jongen, & Van
365 Boekel, 2000). The water lost during processing may result in a concentration of the
366 antioxidant compounds, which could be another reason for the observed increase in the

367 antioxidant activity. The observed differences between the antioxidant activities obtained
368 using both methods might be due to the different principles on which these methods are
369 based. Although both methods are redox-linked colorimetric methods, the DPPH radical
370 assay and FRAP assay are based on the acceptance of either hydrogen atoms or electrons
371 from antioxidants, respectively.

372 4. Conclusions

373 In this study, the antioxidant potential and antioxidant contents, such as vitamin C or
374 phenolic compounds, in *Brassica* vegetables showed significant losses during thermal
375 processing. The results were in line with previous studies where thermal processing
376 resulted in decreased antioxidant activity. This must be considered when calculating the
377 dietary intake of these compounds from cooked vegetables. The cooking conditions
378 evaluated in this study were strong, and smaller losses would be expected under milder
379 cooking conditions. However, this must be confirmed *in vitro*. This study also
380 demonstrated that non-commercial parts of crucifers can be as rich in nutrients as the
381 currently commercially used parts of these plants. For example, the uncooked stems of
382 broccoli could be used as resources for the generation of extracts rich in antioxidants and
383 other value-added ingredients. Future studies would include a sensorial analysis of these
384 underused plant parts and an investigation of their acceptance by consumers. The results
385 obtained herein open new commercial opportunities for *Brassica* producers for their use
386 as novel ingredients in healthy foods. This would not only promote health but also reduce
387 the amount of co-products discarded as waste or used for low-value purposes.

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395 **Figure captions**

396 **Figure 1. Effect of treatment on the vitamin C content of A) Broccoli cv. Marathon;**
397 **B) Broccoli cv. Parthenon; C) Broccoli cv. Graffiti; D) Broccoli cv. Pastoret; E)**
398 **Espigall del Garraf; F) Kale cv. Crispa**

399 Values represent the mean of three independent experiments \pm S.D. Capital letters
400 indicate significant differences between different parts of the same sample. Lower case
401 letters indicate significant differences between treatments. The criterion for statistical
402 significance was $p < 0.05$.

403 **Figure 2. Effect of steaming and *sous-vide* processing on the total phenolic content**
404 **of A) Broccoli cv. Marathon; B) Broccoli cv. Parthenon; C) Broccoli cv. Graffiti; D)**
405 **Broccoli cv. Pastoret; E) Espigall del Garraf; F) Kale cv. Crispa**

406 Values represent the mean of three independent experiments \pm S.D. Capital letters
407 indicate significant differences between different parts of the same sample. Lower case
408 letters indicate significant differences between treatments. The criterion for statistical
409 significance was $p < 0.05$.

410 **Figure 3. Antioxidant activity measured using the FRAP assay of A) Broccoli cv.**
411 **Marathon; B) Broccoli cv. Parthenon; C) Broccoli cv. Graffiti; D) Broccoli cv.**
412 **Pastoret; E) Espigall del Garraf; F) Kale cv. Crispa**

413 Values represent the mean of three independent experiments \pm S.D. Capital letters
414 indicate significant differences between different parts of the same sample. Lower case
415 letters indicate significant differences between treatments. The criterion for statistical
416 significance was $p < 0.05$.

417 **Figure 4. Antioxidant activity measured using DPPH· assay of A) Broccoli cv.**
418 **Marathon; B) Broccoli cv. Parthenon; C) Broccoli cv. Graffiti; D) Broccoli cv.**
419 **Pastoret; E) Espigall del Garraf; F) Kale cv. Crispa**

420 Values represent the mean of three independent experiments \pm S.D. Capital letters
421 indicate significant differences between different parts of the same sample per treatment.
422 Lower case letters indicate significant differences between treatments. The criterion for
423 statistical significance was $p < 0.05$.

424 **Table 1. Colour recordings for the fresh and treated samples. Values represent mean**
 425 **of three independent experiments \pm S.D. The criterion for statistical significance was**
 426 **$p < 0.05$.**

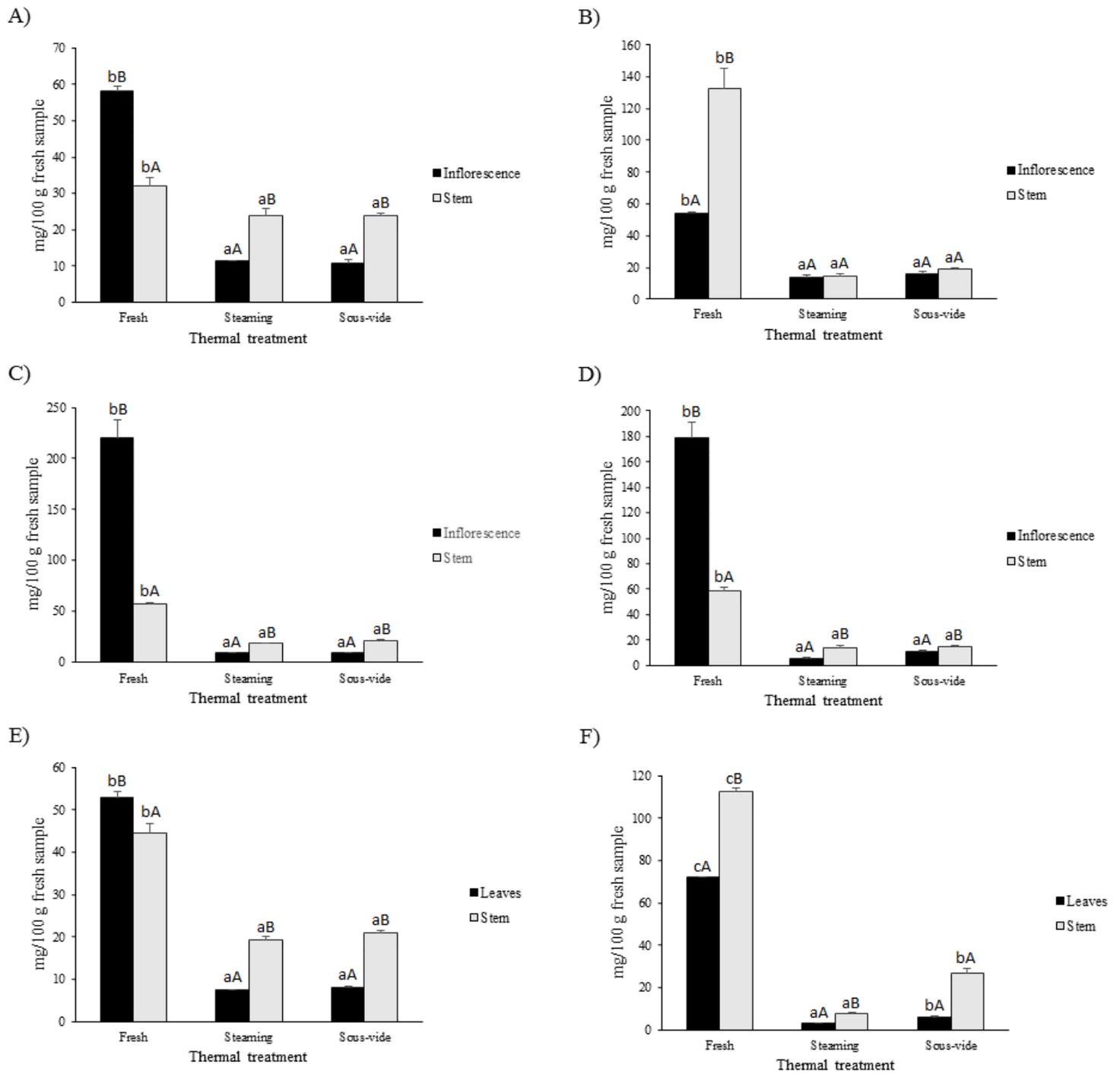
Sample	Part	Treatment	L^*	h^0
Broccoli cv. Marathon	Inflorescence	Fresh	44.14 \pm 4.79 ^{aA}	122.73 \pm 4.29 ^{cA}
		Steaming	40.21 \pm 11.42 ^{aA}	96.92 \pm 2.02 ^{aA}
		<i>Sous-vide</i>	44.88 \pm 16.61 ^{aA}	106.09 \pm 2.22 ^{bA}
	Stalk	Fresh	47.02 \pm 2.18 ^{aA}	122.29 \pm 1.43 ^{aA}
		Steaming	62.23 \pm 3.11 ^{bB}	123.12 \pm 2.07 ^{aB}
		<i>Sous-vide</i>	59.81 \pm 2.48 ^{bA}	127.28 \pm 1.70 ^{bB}
Broccoli cv. Parthenon	Inflorescence	Fresh	49.30 \pm 6.49 ^{aA}	121.47 \pm 2.52 ^{bA}
		Steaming	46.33 \pm 17.37 ^{aA}	101.22 \pm 6.79 ^{aA}
		<i>Sous-vide</i>	47.85 \pm 12.63 ^{aA}	110.86 \pm 4.78 ^{aA}
	Stalk	Fresh	50.24 \pm 3.62 ^{aA}	125.69 \pm 2.01 ^{aA}
		Steaming	60.07 \pm 3.91 ^{bA}	124.28 \pm 5.61 ^{aB}
		<i>Sous-vide</i>	59.78 \pm 2.33 ^{bA}	124.39 \pm 1.23 ^{aB}
Broccoli cv. Graffiti	Inflorescence	Fresh	40.75 \pm 9.64 ^{aA}	111.43 \pm 15.01 ^{aA}
		Steaming	42.95 \pm 3.80 ^{aA}	105.71 \pm 7.21 ^{aA}
		<i>Sous-vide</i>	48.18 \pm 9.08 ^{aA}	106.25 \pm 9.23 ^{aA}
	Stalk	Fresh	45.04 \pm 2.97 ^{aA}	122.87 \pm 1.94 ^{aA}
		Steaming	63.08 \pm 2.20 ^{cB}	122.82 \pm 2.69 ^{aA}
		<i>Sous-vide</i>	56.78 \pm 2.34 ^{bA}	130.55 \pm 4.87 ^{bB}
Broccoli cv. Pastoret	Inflorescence	Fresh	42.64 \pm 7.16 ^{aA}	124.22 \pm 8.90 ^{bA}
		Steaming	42.13 \pm 19.74 ^{aA}	100.35 \pm 5.35 ^{aA}
		<i>Sous-vide</i>	41.31 \pm 13.96 ^{aA}	109.80 \pm 5.76 ^{aA}
	Stalk	Fresh	57.28 \pm 2.53 ^{aB}	115.77 \pm 2.35 ^{aA}
		Steaming	63.99 \pm 4.31 ^{bA}	116.65 \pm 4.38 ^{aB}
		<i>Sous-vide</i>	65.08 \pm 5.00 ^{bB}	120.89 \pm 8.09 ^{aA}
Espigall del Garraf	Leaves	Fresh	44.91 \pm 6.52 ^{aA}	133.42 \pm 3.17 ^{cA}
		Steaming	44.88 \pm 13.20 ^{aA}	105.74 \pm 4.56 ^{aA}
		<i>Sous-vide</i>	36.80 \pm 7.17 ^{aA}	114.40 \pm 3.00 ^{bA}
	Stalk	Fresh	44.79 \pm 1.57 ^{aA}	129.51 \pm 3.96 ^{bA}
		Steaming	61.74 \pm 2.87 ^{bB}	120.69 \pm 2.95 ^{aB}
		<i>Sous-vide</i>	59.52 \pm 2.82 ^{bB}	130.92 \pm 20.28 ^{abA}
Kale cv. Crispa	Leaves	Fresh	37.58 \pm 3.36 ^{cA}	136.12 \pm 2.68 ^{bB}
		Steaming	34.76 \pm 1.51 ^{bA}	110.17 \pm 3.87 ^{aA}
		<i>Sous-vide</i>	30.24 \pm 1.96 ^{aA}	114.52 \pm 3.93 ^{aB}
	Stalk	Fresh	54.38 \pm 1.07 ^{aB}	113.76 \pm 2.80 ^{bA}
		Steaming	59.94 \pm 3.89 ^{bB}	95.13 \pm 1.98 ^{aA}
		<i>Sous-vide</i>	58.70 \pm 1.48 ^{bB}	98.02 \pm 2.27 ^{aA}

Capital letters indicate significant differences between different parts of the same plant for fresh, steamed, or sous-vide processed samples. Lower case letters indicate significant differences between treatments for the same part of the plant.

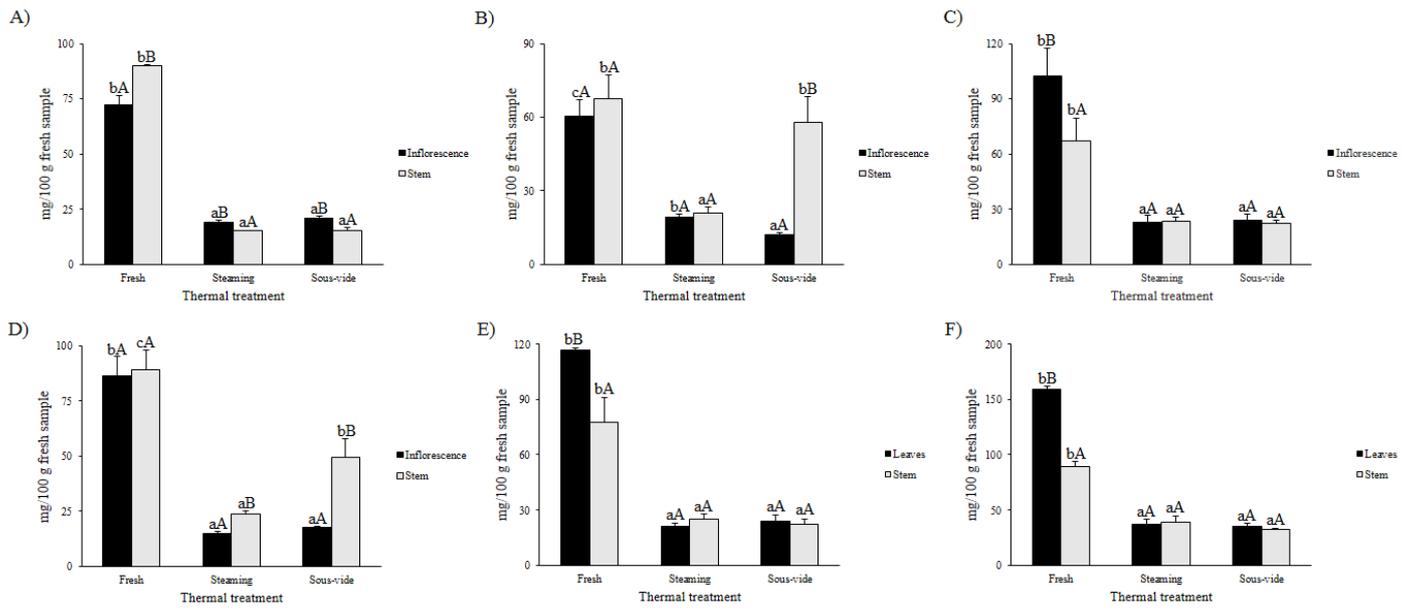
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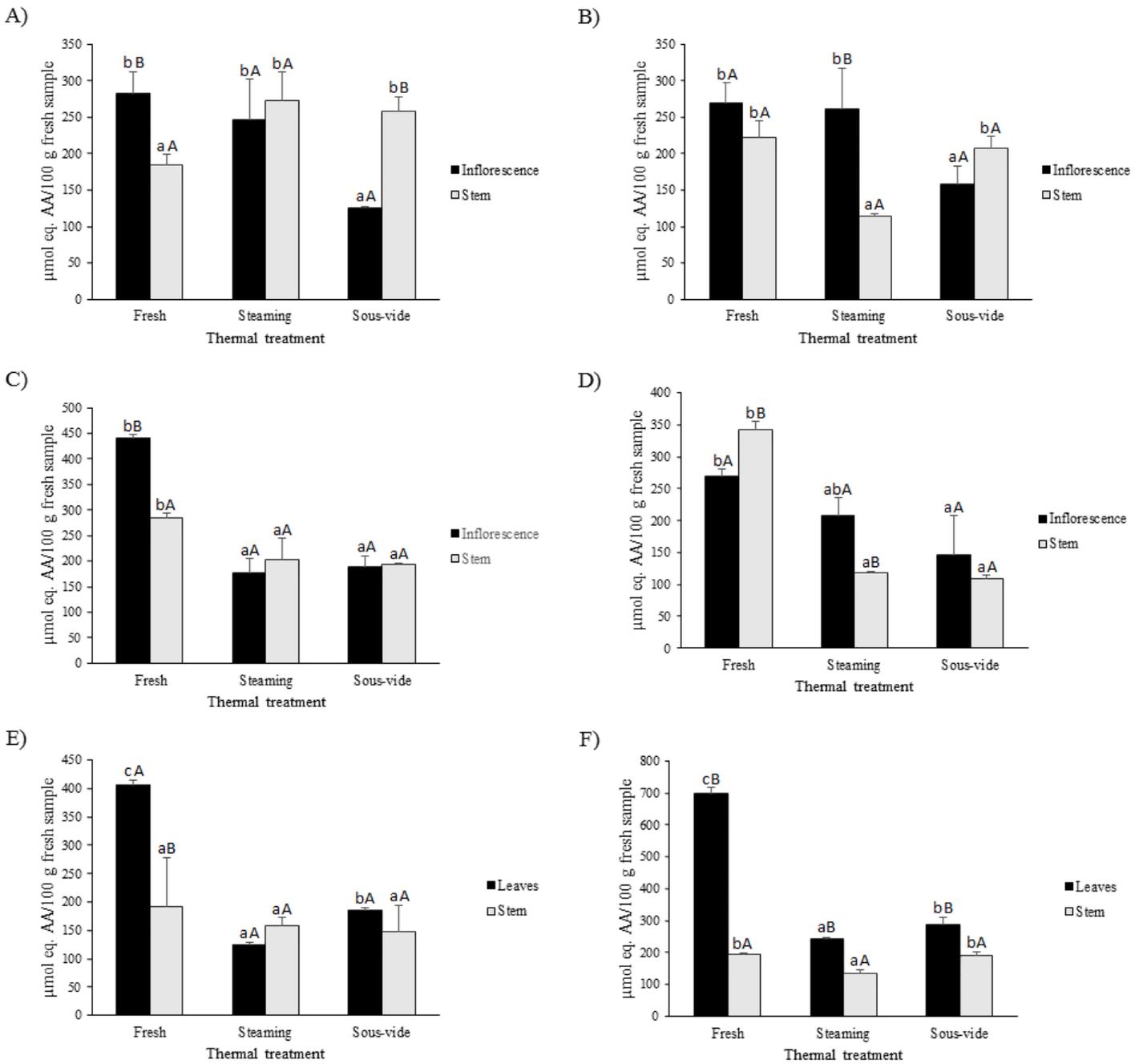
429 **Figure 1**



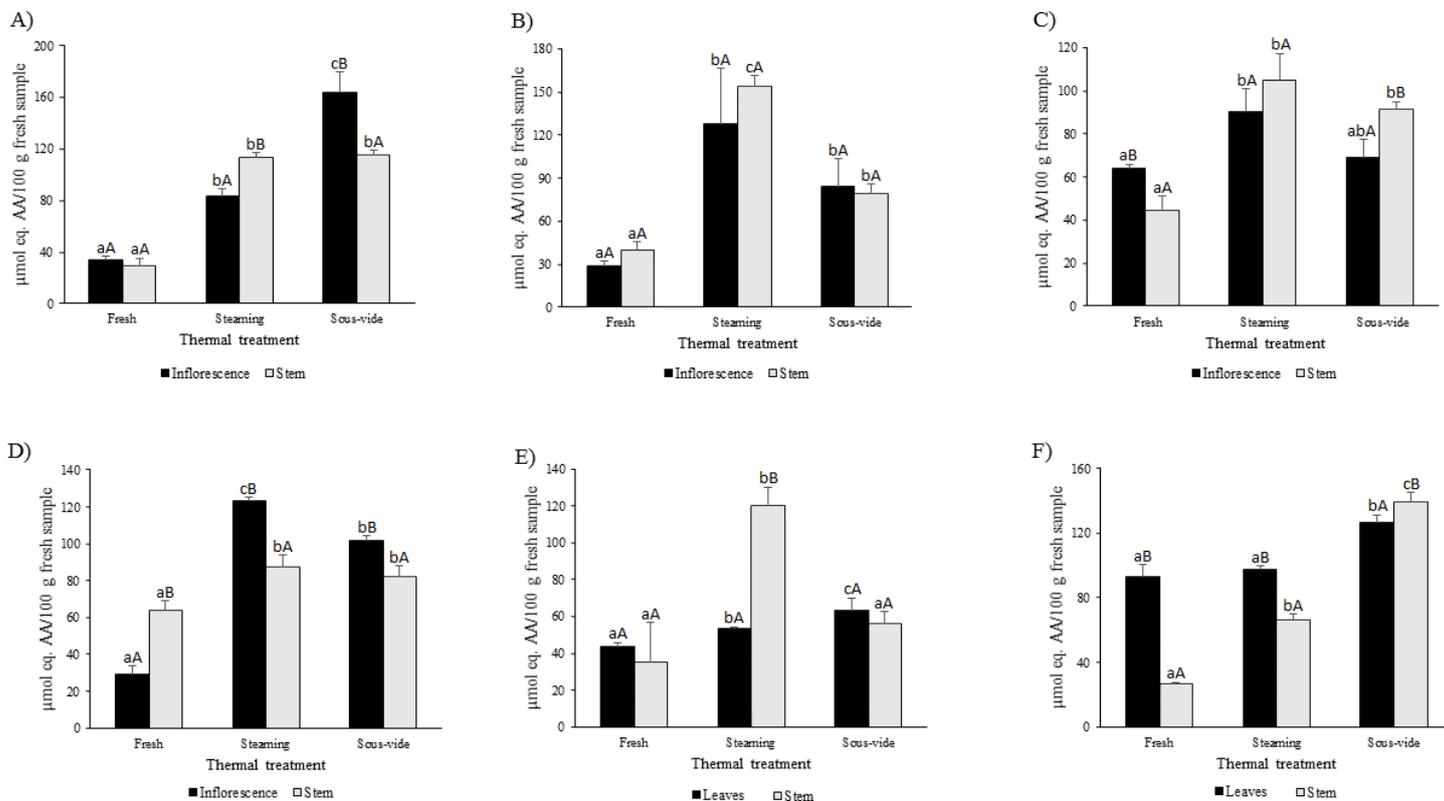
431 **Figure 2**



433 **Figure 3**



434 **Figure 4**



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