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2 **Soil management effects on greenhouse gases production at**
3 **the macroaggregate scale**

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17 **Abstract**

18 Agricultural management practices play an important role in greenhouse gases (GHG)
19 emissions due to their impact on the soil microenvironment. In this study, two
20 experiments were performed to investigate the influence of tillage and N fertilization on
21 GHG production at the macroaggregate scale. In the first experiment, soil
22 macroaggregates collected from a field experiment comparing various soil management
23 systems (CT, conventional tillage; NT, no-tillage) and N fertilization types (a control
24 treatment without N and mineral N and organic N with pig slurry treatments both at 150
25 kg N ha⁻¹) were incubated for 35 days. Methane (CH₄), carbon dioxide (CO₂) and
26 nitrous oxide (N₂O) production was quantified at regular time intervals by gas
27 chromatography. In the second experiment, the effects of fertilization type and soil
28 moisture on the relative importance of nitrification and denitrification processes in N₂O
29 emission from soil macroaggregates were quantified. Nitrate ammonium,
30 macroaggregate-C concentration, macroaggregate water-stability, microbial biomass-C
31 and N (MBC and MBN, respectively) and water-soluble C (WSC) were determined.
32 While NT macroaggregates showed methanotrophic activity, CT macroaggregates acted
33 as net CH₄ producers. However, no significant differences were found between tillage
34 systems on the fluxes and cumulative emissions of CO₂ and N₂O. Greatest cumulative
35 CO₂ emissions, macroaggregate-C concentration and WSC were found in the organic N
36 fertilization treatment and the lowest in the control treatment. Moreover, a tillage and N
37 fertilization interactive effect was found in macroaggregate CO₂ production: while the
38 different types of N fertilizers had no effects on the emission of CO₂ in the NT
39 macroaggregates, a greater CO₂ production in the CT macroaggregates was observed for
40 the organic fertilization treatment compared with the mineral and control treatments.
41 The highest N₂O losses due to nitrification were found in the mineral N treatment while

42 denitrification was the main factor affecting N₂O losses in the organic N treatment. Our
43 results suggest that agricultural management practices such as tillage and N fertilization
44 regulate GHG production in macroaggregates through changes in the proportion of C
45 and N substrates and in microbial activity.

46

47 **Keywords**

48 Carbon dioxide, denitrification, fertilization, macroaggregate, methane, nitrification,
49 nitrous oxide, tillage.

50 **Introduction**

51 The production and consumption of soil greenhouse gases (GHG) is mediated by
52 several microbial processes (Conrad, 1996). For instance, soil carbon dioxide (CO₂)
53 emissions are the result of microbial heterotrophic respiration while methane (CH₄) is
54 normally oxidized by methanotrophic bacteria in aerobic soils (Goulding et al. 1995).
55 Furthermore, soil nitrous oxide (N₂O) production is the result of nitrification and
56 denitrification processes (Blackmer et al., 1980; Firestone et al., 1980; Poth and Focht,
57 1985). Those microbial processes are regulated by the physical protective capacity of
58 aggregates that limit decomposition of organic C and N compounds (Elliott, 1986). Soil
59 aggregates not only protect C and N, but they also regulate both the structure and the
60 activity of the soil microbial community (Gupta and Germida, 1988; Miller et al., 2009).
61 The intra-aggregate distribution of pores plays a major role in microbial access to
62 oxygen, substrates and water. As Young and Ritz (2000) pointed out, soil structure
63 regulates oxygen diffusion to habitat sites, depending on the connectivity and tortuosity
64 of pore pathways. The aggregate architecture also controls the distribution of water
65 films within soil matrix, affecting microbial microhabitats. Thus, the diffusion of
66 oxygen to the center of aggregates will depend on the spatial arrangement of water films
67 (Young and Ritz, 2000). The last factors affect the importance of denitrification and
68 respiration activities and demonstrate the role played by soil aggregates regulating them
69 (Beare et al., 1994; Estavillo et al., 2002). Moreover, due to their physical protective
70 capacity, soil aggregates also regulate the microbial accessibility to substrates.

71 In a recent experiment, Lenka and Lal (2013) have suggested that the aggregate
72 hierarchy theory of Tisdall and Oades (1982) could be extended to describe the effect of
73 soil aggregation on GHG emission from soil. That theory postulates that the nature of
74 the organic binding agents (transient, temporary and persistent) regulates different

75 hierarchical stages of aggregation. Microaggregates are formed by the joining of
76 primary particles and silt-sized aggregates and persistent organic binding agents, while
77 these microaggregates are bound together into macroaggregates by temporary and
78 transient organic binding agents. These organic materials are protected by the
79 heterogeneity of the soil microenvironment which limits the access of decomposers and
80 their enzymes (Schmidt et al., 2011; Ananyeva et al., 2013).

81 The agricultural practices play an important role in GHG emissions due to their effects
82 on the soil microenvironment. Tillage breaks soil aggregates leading to enhanced
83 organic matter decomposition (Álvaro-Fuentes et al., 2008; Beare et al., 1994) and
84 reduced C and N concentration (Plaza-Bonilla et al., 2010). Contrarily, the use and
85 maintenance of no-tillage (NT) increases the stability of soil macroaggregates (Plaza-
86 Bonilla et al., 2013b), a fact that could lead to a reduction in heterotrophic respiration
87 due to a greater substrate protection, thus limiting the emissions of CO₂. Likewise, CH₄
88 production is also affected by tillage management. For instance, in a wheat-fallow
89 rotation, Kessavalou et al. (1998) reported higher CH₄ uptake rates under NT when
90 compared with a plough treatment. Also, Hütsch (1998a) reported 4.5-11 times greater
91 CH₄ oxidation rates under NT than under conventional tillage (CT). Ball et al. (1999)
92 hypothesized that the reduction in CH₄ oxidation usually found when tillage is
93 performed could be due to the disturbance of the methanotrophic microbes by tillage,
94 the changes in gas diffusivity or a long-term damage to methanotrophs due to disruption
95 of soil structure. Tillage also has an impact on N₂O emissions. Estavillo et al. (2002),
96 studying the effects of ploughing a permanent pasture on the emissions of this gas,
97 observed an increase in both soil organic N mineralization and N₂O production rates
98 from nitrification and denitrification processes after the breakage of soil aggregates by
99 tillage.

100 Nitrogen fertilization has a strong impact on soil aggregation and C and N protection.
101 The application of organic fertilizers such as pig slurry enhances the proportion of
102 easily-decomposable C fractions (Morvan and Nicolardot, 2009) that could act as
103 substrates for the denitrification process and the concomitant soil N₂O emissions to the
104 atmosphere (Burford and Bremner, 1975). Sexstone et al. (1985) quantified the
105 diffusion of oxygen within soil aggregates establishing a relationship between their size
106 and their potential to act as denitrifying microsites within soil. Nitrogen fertilization
107 also plays a major role in methane oxidation. Different authors (Hütsch et al. 1993;
108 Mosier et al. 1991; Steudler et al. 1989), working with incubated soil cores from
109 agricultural, grassland and forest experiments, observed a decrease in CH₄ uptake when
110 applying inorganic N to soil. Contrarily, recent findings suggest that ammonium-based
111 fertilizers could stimulate the activity of methanotrophs (Bodelier and Landbroek,
112 2004).

113 In recent years, different experiments have been performed to analyze the effects of
114 aggregate size on CO₂, CH₄ and N₂O production (Diba et al., 2011; Drury et al., 2004;
115 Kimura et al., 2012). However, inconsistent results have been observed in the literature
116 due to the simultaneous diverse microbial processes that soil aggregates can hold (Sey
117 et al., 2008). For instance, Parkin (1987) related the spatial heterogeneity in the N₂O
118 emissions usually observed in most experiments with the presence of particulate organic
119 matter within soil aggregates. Those studies demonstrate that different aggregate
120 attributes such as size or C fractions within them regulate GHG production processes.
121 However, few experiments have studied the effects of agricultural management
122 practices on soil GHG production at the aggregate scale.

123 Thus, the objectives of this study were: (i) to analyze the effect of the use of different
124 types of tillage and N fertilization on the production of GHG by soil macroaggregates

125 and, (ii) to quantify the relative importance of the nitrification and denitrification
126 processes on the macroaggregate emissions of N₂O depending on the type of fertilizer
127 used. We hypothesized that (i) CT macroaggregates would emit a greater amount of
128 GHG due to their lower protection of the organic C and N compounds when compared
129 to NT macroaggregates and (ii) the application of pig slurry and mineral N would result
130 in different rates of GHG production provided by soil macroaggregates.

131 **Materials and Methods**

132 Soil samples were collected from an experimental field established in 2010 in Senés de
133 Alcubierre, NE Spain (41° 54' 12'' N, 0° 30' 15'' W), in an area with a temperate
134 continental Mediterranean climate. This field experiment has a randomized block design
135 with three replications comparing different tillage systems and N fertilization
136 treatments. Two tillage systems (CT, conventional tillage with disk ploughing and NT,
137 no-tillage) and two types of N fertilizers (mineral N with ammonium nitrate and
138 ammonium sulphate and organic N with pig slurry), with three N doses (0, 75 and 150
139 kg N ha⁻¹), were compared. Each year, in the CT treatment, tillage is performed right
140 before the seeding of barley (*Hordeum vulgare L.*) with one pass of a disk plough to 20
141 cm depth in October, after the application of organic and mineral fertilizers. The NT
142 treatment consisted of a total herbicide application (1.5 L 36% glyphosate per hectare)
143 for controlling weeds before sowing. Mineral N fertilizer was manually applied. The
144 treatment with 150 kg N ha⁻¹ was split into two applications: half of the dose before
145 tillage as ammonium sulphate (21% N) and the other half at the beginning of tillering,
146 in February, as ammonium nitrate (33.5% N). For the 75 kg N ha⁻¹ treatment the entire
147 dose was applied at tillering as ammonium nitrate. Equally, in the treatments with
148 organic fertilization, the 75 kg N ha⁻¹ rate was applied entirely at tillering and the 150 kg
149 N ha⁻¹ one was split into two applications, one half before tillage and the other half at
150 tillering. The organic fertilization treatment consisted of the application of pig (*Sus*
151 *scrofa*) slurry from a commercial farm in the area. The slurry was conventionally
152 surface-spread using a commercial vacuum tanker fitted with a splashplate. The
153 machinery was previously calibrated to apply the precise dose after analyzing the pig
154 slurry. The main edaphoclimatic characteristics of the experimental site are listed in
155 Table 1. Prior to the establishment of the experiment the field was conventionally tilled

156 and fertilized with mineral N for four decades until 2008. Then, the whole field was
157 transformed to no-tillage. Finally, as commented before, when the experiment started in
158 2010, the CT plots were added. The cropping system is a continuous barley
159 monoculture.

160 ***Experiment 1: GHG production from soil macroaggregates under different tillage***
161 ***and N fertilization treatments.***

162 Soil samples were obtained from both tillage treatments (CT and NT) and the lower (0
163 kg N ha⁻¹, Control) and the higher (150 kg mineral N ha⁻¹, Mineral, and 150 kg organic
164 N ha⁻¹ with pig slurry, Organic) fertilization treatments of the field experiment. Soil
165 sampling was performed in March 2012 during the late tillering stage of the crop, three
166 weeks after the top-dressing application of fertilizers. In each plot (i.e., tillage and N
167 fertilization treatments), soil samples were collected from four areas that correspond to
168 the four replications. From each sampling area, a composite sample of approximately
169 500 g was taken from the 0-5 cm soil depth using a flat spade and outside the wheel
170 tracks areas. Afterwards, the samples were stored in crush-resistant airtight plastic
171 containers for 3-4 hours. Once in the laboratory, the samples were gently passed
172 through an 8mm sieve and air-dried at room temperature. Soil macroaggregates (0.250-
173 8 mm) were obtained placing the 8-mm soil sieved sample on the top of a 0.250 mm
174 sieve in an electromagnetic sieve apparatus (Filtrá FTL-0200, Badalona, Spain). A
175 sieving time of 1 min and the lowest power program of the device were used to avoid
176 macroaggregate breakage. The dry-sieved macroaggregates (0.250-8 mm) obtained
177 were stored in aluminium trays taking care to avoid any breakage until further analyses.

178 Samples of 40-g each of dry-sieved macroaggregates (0.250-8 mm) were placed in 500
179 ml Mason jars. Four jars were built for each tillage and N fertilization combination. A

180 stainless steel fitting turned to accommodate two silicon-Teflon septa was inserted in
181 the lid of each jar to ensure air tightness. A volume of 12.8 ml of distilled water was
182 added to each macroaggregate sample using a micropipette in order to avoid the
183 breakage of the macroaggregates when adding the water, and also to obtain a
184 gravimetric moisture content of about 32%. This value corresponds to the field capacity
185 of the bulk soil of our experiment according to Saxton and Rawls (2006). All the jars
186 were covered with a layer of parafilm which was pinpricked to ensure air exchange and
187 avoid sample desiccation during the incubation process. The weight of the jars with the
188 wet macroaggregate samples was recorded and then every 48 hours to check for water
189 evaporation. Distilled water was added when needed. Air samples were withdrawn at
190 0(0), 4(0.17), 12(0.5), 24(1), 48(2), 72(3), 192(8), 384(16), 504(21), 672(28) and
191 840(35) hours(days) after the beginning of the incubation process. The parafilm layer of
192 each jar was removed 15 min prior to each gas sampling. Then all the lids were tightly
193 closed and a 15 ml headspace gas sample was withdrawn with the use of a gas-tight
194 syringe, pumping twice before the extraction to ensure a total mixing of the gas in the
195 jar (0 min sampling). Afterwards, 15 ml of ambient air were injected in the jars to
196 compensate for the volume previously withdrawn. A second gas sampling was
197 performed 60 min later. The gas samples obtained were injected in 12 ml Exetainer
198 borosilicate glass vials (model 038W, Labco, High Wycombe, UK) until their analysis.
199 Once the samplings were made (i.e., after 60 min) the lids were opened and the jars
200 were covered with a parafilm layer until the next sampling event. Also, the jars were
201 covered during the incubation to avoid light exposure. As explained in the next section,
202 the difference in GHG concentration between 0 and 60 min samplings was used to
203 calculate the GHG fluxes.

204 *Gas and soil analysis*

205 The gas samples were analyzed with an Agilent 7890A gas chromatography system
206 equipped with an electron capture detector (ECD) and a flame ionization detector (FID)
207 plus methanizer, and three automated valves to obtain the three gases of interest (i.e.,
208 CH₄, CO₂ and N₂O) for each gas sample injection. A HP-Plot Q column (30 m long,
209 0.32 mm of section and 20 μm) was used along with a 15 m long pre-column of the
210 same material. The injector and the oven temperature were set to 50°C. The temperature
211 of the FID and ECD detectors was set to 250°C and 300°C, respectively. The methanizer
212 temperature was set to 375°C. For the FID detector, H₂ was used as a carrier gas and N₂
213 as a make-up gas at 35 and 25 ml min⁻¹, respectively. In the case of the ECD detector,
214 5% methane in Argon was used as a make-up gas at 30 ml min⁻¹. The volume of sample
215 injected was 1 ml. The system was calibrated using analytical grade standards (Carbueros
216 Metálicos, Barcelona, Spain). Soil CH₄, CO₂ and N₂O production in the jar headspace
217 was calculated according to Holland et al. (1999). Gas concentrations (ppm) obtained
218 with the chromatography system were converted to mass units with the ideal gas
219 equation:

$$220 \quad C_m = (C_v \times M \times P) / (R \times T)$$

221 where C_m is the mass/volume concentration (e.g., mg CO₂-C m⁻³ incubation jar
222 headspace), C_v is the volume/volume concentration (ppm of each GHG obtained with
223 the chromatography system), M is the molecular weight of each GHG (e.g., 12 g CO₂-C
224 mol⁻¹ or 28 g N₂O-N mol⁻¹), P is atmospheric pressure, R is the universal gas constant
225 and T is the incubation temperature (298 K). C_m was multiplied by the headspace
226 volume of the incubation jars (5 x 10⁻⁴ m³) to obtain the mass of CH₄-C, CO₂-C or
227 N₂O-N accumulated during the incubation. Thus, the mass of GHG produced (e.g., mg
228 CH₄-C kg⁻¹ macroaggregates h⁻¹) is calculated as follows:

229 $f = ((C_1 - C_0) / (m \times t)) \times 1000$

230 where f is the mass of gas produced per unit of time, C_1 and C_0 are the mass of C or N
231 produced at the end and at the beginning of two consecutive samplings, respectively, m
232 is the mass of air-dried macroaggregates in each jar (0.04 kg) and t is the incubation
233 period (1 h). Finally, the cumulative production of $\text{CH}_4\text{-C}$, $\text{CO}_2\text{-C}$ and $\text{N}_2\text{O-N}$ was
234 calculated using the trapezoid rule by linear interpolation between two consecutive
235 samplings.

236 Additionally, the initial mineral N (i.e., nitrate and ammonia), the C concentration, and
237 the proportion of water-stable macroaggregates were quantified for each experimental
238 unit. Once the incubation was finished, the microbial biomass-C and microbial biomass-
239 N (MBC and MBN, respectively), the nitrate and ammonia content, the water-soluble C
240 (WSC) and the C concentration of each 40 g macroaggregates sample were also
241 quantified. Soil nitrate (NO_3^-) and ammonium (NH_4^+) were determined extracting 10 g
242 of macroaggregates with 80 ml of 1 M KCl and using a continuous flow autoanalyzer
243 (Seal Autoanalyzer 3). The macroaggregate-C concentration was quantified by the wet
244 oxidation method of Walkley-Black described by Nelson and Sommers (1996), with a
245 modification to increase the digestion of soil organic carbon (SOC), which consisted in
246 boiling the sample and the extraction solution at 150°C for 30 min (Mebius, 1960). The
247 proportion of water-stable macroaggregates and their sand content were determined
248 following a modification of the method of Elliott (1986) as described in Plaza-Bonilla et
249 al. (2013a). The microbial biomass-C and microbial biomass-N were determined with
250 the chloroform-fumigation and direct extraction method of Vance et al. (1987). The
251 extracts were analyzed for organic C and N with a multi C/N TOC-TNB analyzer 3100
252 (Analytik Jena, Jena, Germany). The extraction coefficient applied for both C and N
253 was 0.38 (Sparling and Zhu, 1993; Vance et al., 1987). The WSC was extracted by

254 shaking 10 g of macroaggregates in 40 ml of distilled water with 0.5 g potassium
255 sulphate in a centrifuge tube for 30 min, centrifuging for 5 min at 5000 rpm and filtering
256 all supernatant solution through a Whatman no.42 filter. The organic C in the filtrate
257 was determined by the same device used for the MBC-MBN determination.

258 *Data analysis*

259 Cumulative GHG data were log-transformed and analyzed using the SAS statistical
260 software (SAS Institute Inc., 1990). To compare the effects of tillage, fertilizer
261 treatments and sampling time on cumulative GHG emissions, a repeated measures
262 analysis of variance for a bifactorial design was performed for each gas. When
263 significant, differences among treatments were identified at the 0.05 probability level of
264 significance using an LSD test. For each sampling time, the linear relationship between
265 CO₂ and N₂O production in the CT and NT macroaggregates was determined with the
266 statistical package JMP 10 (SAS Institute Inc, 2012). To analyse the relationship
267 between CO₂ production, proportion of water-stable macroaggregates and their C
268 concentration, a stepwise regression was performed using the statistical package JMP
269 10 (SAS Institute Inc, 2012).

270

271 ***Experiment 2: Relative importance of nitrification and denitrification in N₂O*** 272 ***production from soil macroaggregates under different N fertilization types.***

273 Soil samples from the 0-5 cm soil depth were collected in the same fertilization
274 treatments as in Experiment 1 (i.e., 0 and 150 kg N ha⁻¹ as mineral N and pig slurry),
275 only under NT. However, for this experiment soil sampling was performed on
276 December 2012, three weeks after the pre-seeding fertilization of the crop. In each plot

277 (i.e., N treatment), six areas that would correspond to the six replications of the
278 experiment were defined.

279 From each area, a 500 g composite soil sample was taken from the 0-5 cm soil depth
280 using a flat spade, taking care to avoid the wheel track areas. Dry-sieved
281 macroaggregates fractionation was analogous to Experiment 1. The experimental set up
282 consisted of three N fertilization types (0 kg N ha⁻¹, mineral N at 150 kg ha⁻¹ and
283 organic N with pig slurry at 150 kg ha⁻¹), two soil moisture treatments (15% and 30 %
284 gravimetric water content) and three levels of acetylene (0%, 0.01% and 5%, v v⁻¹).
285 Each combination of the three factors was repeated six times according to the
286 experimental replications. Therefore, the total number of observations was 108. To
287 achieve this number of observations, the dry-sieved macroaggregates from each
288 experimental replication was divided in six subsamples of 40 g that were placed in
289 Mason jars. These six subsamples were divided in two groups. In the first group, three
290 subsamples were moistened with distilled water to 15% gravimetric water content. The
291 other three subsamples were moistened to 30% gravimetric water content. The lids of
292 the jars were closed and a 15 ml headspace gas sample was taken for every jar (0 min
293 sampling). Afterwards, for each soil moisture treatment three acetylene (C₂H₂)
294 treatments were applied: 0%, 0.01% and 5% (v v⁻¹) corresponding to partial pressures of
295 0 Pa, 10 Pa and 5000 Pa, respectively) following the method proposed by Klemetsson
296 et al. (1988) to differentiate the relative contribution of the nitrification and
297 denitrification processes in N₂O emissions. Different drawbacks of the method have
298 been reported in the literature. Among them, Baggs (2008) enumerates (i) a possible
299 underestimation of denitrification by preventing the supply of nitrifier-NO₃⁻, mainly in
300 aquatic systems (Groffman et al., 2006), (ii) acetylene could be used as a C-substrate for
301 denitrification, and (iii) a limited diffusion of acetylene into fine textured soils.

302 However, acetylene-based methods still have a role in systems with high NO_3^-
303 concentrations (Groffman et al., 2006), such as the agricultural soil of our experiment,
304 and are useful for comparative purposes between different treatments (Estavillo et al.,
305 2002). According to each treatment, different volumes of ambient air were injected to
306 equilibrate the pressure into the jars. The jars with the macroaggregates were incubated
307 at 25°C for 24 hours. After that, another 15 ml gas sample was withdrawn to calculate
308 the accumulation of N_2O in the 24 hours period for each jar. Air gas samples were
309 stored and analyzed following the same methodology as in Experiment 1. It was
310 assumed that the N_2O measured in the treatment without acetylene (i.e., 0% C_2H_2)
311 corresponded to the N_2O produced by the nitrification and denitrification processes. In
312 turn, the N_2O measured in the treatment with 0.01% C_2H_2 corresponded only to that
313 produced during the denitrification process (Davidson et al., 1986) and, finally, the N_2O
314 measured in the treatment with a C_2H_2 concentration of 5% corresponded to the N_2O
315 produced due to a complete denitrification (Yoshinari et al., 1977). The production of
316 N_2O by the nitrification process was calculated from the difference between the N_2O
317 measured in the 0% and the 0.01% C_2H_2 treatments, while the production of N_2O by the
318 denitrification process corresponded to the amount of N_2O measured in the 0.01% C_2H_2
319 treatment, and complete denitrification (i.e., N_2O that would be reduced to N_2) was
320 calculated as the difference between the N_2O measured in the 5% and the 0.01% C_2H_2
321 treatments. The gas samples were analyzed with an Agilent 7890A gas chromatography
322 system equipped with an ECD detector with the same parameters as in Experiment 1.
323 Moreover, the mineral N content as nitrate and ammonium and the WSC were also
324 determined prior to the incubation following the methodology described above.

325 *Data analysis*

326 The N₂O production data were transformed using the Box-Cox procedure and analyzed
327 using the SAS statistical software (SAS Institute Inc., 1990). To compare the effects of
328 fertilizer treatments and soil moisture on N₂O production an analysis of variance was
329 performed. When significant, differences among treatments were identified at the 0.1
330 probability level of significance using an LSD test. Furthermore, the linear relationship
331 between WSC and N₂O production was determined with the statistical package JMP 10
332 (SAS Institute Inc., 2012)

333 **Results**

334 *Experiment 1: GHG production from soil macroaggregates under different tillage and*
335 *N fertilization treatments.*

336 Tillage significantly affected the fluxes of CH₄ produced by soil macroaggregates. As an
337 average of all the samplings performed during the incubation period, macroaggregates
338 of the CT treatment acted as emitters of CH₄ while those under NT acted as a CH₄ sink
339 (Table 2). Also, significant differences on cumulative CH₄ fluxes were observed
340 between CT and NT (Fig. 1a). According to the data, the methanotrophic activity in the
341 NT treatment began after the first 72 hours of macroaggregate incubation (Fig. 1a). In
342 contrast to CH₄, no significant differences were found between tillage systems on
343 neither the fluxes nor the cumulative emissions of CO₂ and N₂O (Table 2, Fig. 1b and
344 c).

345 Nitrogen fertilization treatments did not affect the fluxes of CH₄ and N₂O (Table 2).
346 Also, cumulative emissions of CH₄ and N₂O did not differ between N fertilization
347 treatments (Fig. 2a and c). CO₂ followed a different trend, with greater average fluxes in
348 the organic treatment (1669.4 μg CO₂-C kg macroaggregates⁻¹ h⁻¹) when compared with
349 the control (1217.5 μg CO₂-C kg macroaggregates⁻¹ h⁻¹) and the mineral (1199.4 μg
350 CO₂-C kg macroaggregates⁻¹ h⁻¹) treatments (Table 2). Also, cumulative CO₂ emissions
351 were the greatest under the organic fertilization treatment in the first 48 hours of the
352 incubation, without differences between the control and mineral treatments (Fig 2b).
353 When the incubation was finished (i.e., after 840 hours), the organic treatment presented
354 a greater cumulative CO₂ emission when compared with the mineral treatment, while
355 the control presented intermediate values (Fig. 2b).

356 The interaction between tillage and N fertilization significantly affected the fluxes of
357 CO₂ (Table 2). The different N fertilization treatments did not show different CO₂
358 fluxes for the NT macroaggregates, whereas the CT macroaggregates under organic
359 fertilization emitted greater amount of CO₂ (1824.5 μg CO₂-C kg macroaggregates⁻¹ h⁻¹)
360 compared with the control (1021.4 μg CO₂-C kg macroaggregates⁻¹ h⁻¹) and mineral
361 (1155.1 μg CO₂-C kg macroaggregates⁻¹ h⁻¹) fertilization treatments (Table 2).

362 No differences between tillage systems were found in the organic carbon (OC)
363 concentration of dry-sieved macroaggregates before or after 840 hour incubation (Table
364 3). However, different results arose when analyzing the OC concentration in the soil
365 macroaggregates under different N fertilization treatments. In this case, greater
366 macroaggregate-C concentration was found in the organic fertilization treatment both
367 before and after the incubation period when compared with the control and mineral
368 treatments (Table 3). Nevertheless, the decrease in the OC concentration during the
369 incubation was not statistically different between N fertilization treatments. Significant
370 differences between tillage and N fertilization treatments were found on the initial NO₃⁻
371 concentration of the macroaggregates (Table 3). A greater initial NO₃⁻ concentration
372 was found in the CT treatment than in the NT treatment, with 110.4 and 92.6 mg NO₃⁻-
373 N kg⁻¹ dry-sieved macroaggregates, respectively. In the case of N fertilization, the
374 mineral treatment showed the greatest initial NO₃⁻ concentration while the control
375 presented the smallest one and the organic treatment intermediate values. After the
376 incubation period (i.e., 840 hours) the mineral and organic fertilization treatments
377 showed greater NO₃⁻ concentration when compared with the control. Also, a significant
378 interaction between tillage and N fertilization was found on this variable: while the
379 macroaggregates of the control and organic fertilization treatments presented no
380 significant differences between CT and NT, the macroaggregates of the CT treatment

381 fertilized with mineral N presented a greater amount of initial NO_3^- when compared to
382 the ones of the NT treatment (Table 3). Significant differences between tillage and N
383 fertilization treatments were also found on the NO_3^- concentration variation (0 vs. 840
384 hours). In this case, the NT-control treatment presented the greatest increase in the NO_3^-
385 concentration in the macroaggregates, followed by the CT-control treatment (Table 3).

386 Differences between N fertilization treatments were also found on the initial NH_4^+ -N
387 concentration and its variation during the incubation process. The organic treatment
388 presented the greatest values, followed by the mineral and the control treatments (Table
389 3). The reduction of the NH_4^+ -N concentration during the incubation period was higher
390 in the fertilized treatments (about 87% and 93% reduction in the NH_4^+ concentration in
391 the mineral and the organic treatments, respectively) compared with the control (about
392 32% reduction) (Table 3). Furthermore, no differences between treatments were found
393 on the MBC content. However, a greater MBN content was found in the organic
394 treatment compared with the mineral and the control treatments (Table 3). The WSC
395 content after the incubation process was significantly affected by both tillage and N
396 fertilization treatments. Thus, a greater WSC content was found under NT than under
397 CT and in the organic N treatment compared with the mineral and control ones (Table
398 3).

399 A highly significant polynomial relationship ($r^2 = 0.72$; $p < 0.001$) was observed between
400 the initial NH_4^+ concentration in the macroaggregates and the cumulative N_2O -N
401 emission during the first 48 hours of incubation (Fig. 3). Furthermore, significant linear
402 relationships were observed between CO_2 and N_2O production in six samplings in CT
403 and in three samplings in NT (Table 4).

404 At the end of the incubation period, a greater proportion of water-stable
405 macroaggregates was quantified under NT compared with CT (Fig. 4). Moreover, a
406 significant interaction ($P < 0.05$) between tillage and N fertilization was found on the
407 water-stability of macroaggregates. While under NT no differences between fertilization
408 treatments were observed in the proportion of water-stable macroaggregates, under CT a
409 greater proportion of water-stable aggregates was found in the organic treatment when
410 compared with the mineral treatment, with intermediate values in the control (Fig. 4).

411 *Experiment 2: Relative importance of nitrification and denitrification on N₂O*
412 *production from soil macroaggregates under different N fertilization types.*

413 No differences between N fertilization treatments were found on NH_4^+ concentration of
414 the macroaggregates before the incubation (Table 5). In contrast, before the incubation
415 process, the fertilized treatments (mineral and organic) presented a greater NO_3^-
416 concentration in the macroaggregates when compared with the control (Table 5).
417 Moreover, significant differences between N fertilization treatments were found on the
418 WSC concentration with greater values in the organic fertilization treatment when
419 compared with the mineral and the control ones (Table 5).

420 At the 15% moisture level, total losses of N as N_2O and N_2 during the incubation of the
421 macroaggregates resulted in 173, 254 and 139 $\text{mg N kg}^{-1} \text{h}^{-1}$ for the control, mineral and
422 organic treatments, respectively (Table 6). In turn, at the 30% moisture level, N losses
423 reached 4751, 5552 and 4922 $\text{mg N kg}^{-1} \text{h}^{-1}$ for the control, mineral and organic
424 treatments, respectively (Table 6). Both N fertilization and soil moisture content
425 significantly affected the amount of $\text{N}_2\text{O-N}$ produced due to the nitrification and
426 denitrification processes (Table 6). The production of N_2O due to the nitrification
427 process was greater in the mineral treatment when compared with the control, with

428 intermediate values in the organic treatment. Different results were obtained in the
429 production of N_2O due to the denitrification process. In this case, the organic treatment
430 showed greater values than the control, while intermediate values were found in the
431 mineral treatment (Table 6). Nevertheless, the production of N_2 due to a complete
432 denitrification process was only affected by soil moisture, with the greatest values in the
433 30% moisture treatment when compared with the 15% moisture treatment (Table 6).
434 Moreover, the production of N_2O due to nitrification and denitrification processes was
435 4.3 and 7.3 times greater in the 30% than in the 15% moisture treatment, respectively
436 (Table 6). No significant relationship was found between WSC and the amount of N_2O
437 produced during the denitrification process (data not shown).

438 **Discussion**

439 *Effects of tillage and N fertilization on GHG production from soil macroaggregates*

440 CH₄ was the only greenhouse gas produced by the macroaggregates that presented
441 significant differences between tillage treatments. In the CT treatment macroaggregates
442 acted as CH₄ producers, whereas macroaggregates of the NT treatment oxidized CH₄
443 mainly from the first 72 hours until the end of the incubation. Methanotrophic activity is
444 reduced by anoxic conditions. In our experiment, an equal amount of water was added
445 to the macroaggregates of both tillage treatments to bring them to the field capacity of
446 undisturbed soil in our field experiment. Therefore, it could be hypothesized that
447 differences in the intra-aggregate pore architecture and connectivity could have
448 maintained a higher amount of aerobic microsites within the NT macroaggregates, thus
449 facilitating the oxidation of CH₄. This hypothesis is in line with the findings of
450 Kravchenko et al. (2013) who studied the effects of tillage on the intra-aggregate
451 porosity of macroaggregates and observed higher intra-aggregate porosity >100 µm in
452 NT macroaggregates when compared with CT macroaggregates. Another hypothesis
453 could be the influence of the different types of tillage on the diversity of
454 microorganisms within macroaggregates, which could have maintained a greater
455 amount of methanotrophs in the NT treatment.

456 According to our results, no differences between tillage systems were found on the
457 amount of macroaggregate-C mineralized as CO₂. That result could be related to the
458 absence of differences in macroaggregate-C concentration prior and after the
459 incubation. Different results were obtained by Fernández et al. (2010) when using soil
460 of a long-term (14-yr) experiment. These authors observed higher production of CO₂ by
461 NT macroaggregates when compared with CT macroaggregates and related this finding

462 to the higher amount of organic C in the NT macroaggregates. Thus, in our experiment,
463 the similar macroaggregate-C concentration found in the CT and NT treatments would
464 have influenced the lack of differences in CO₂ production by macroaggregates.

465 Similarly to CO₂ production, no differences between tillage treatments were found on
466 the fluxes and cumulative emissions of N₂O by soil macroaggregates. Although a higher
467 initial NO₃⁻ concentration susceptible to denitrification and a greater reduction in the
468 concentration of NH₄⁺ during the incubation were found under CT, no greater MBN
469 content was found in this treatment when compared with NT. For that reason, the
470 hypothesis of a greater N immobilization under CT was not supported by our data.
471 Another hypothesis could be a greater or complete denitrification under CT, in which
472 the mineral N would be emitted as N₂. It is known that the N₂/(N₂ + N₂O) ratio
473 increases with decreasing O₂ concentration (Tiedje, 1988). That hypothesis would be in
474 line with the smaller intra-aggregate porosity described above and the related higher
475 anaerobic conditions in CT macroaggregates.

476 Wrage et al. (2001) suggested that a greater soil organic matter content and better
477 aggregate structure could facilitate O₂ diffusion, thus reducing the production of N₂O in
478 NT soils. In our experiment, we found a greater water-stability of macroaggregates
479 under the NT treatment when compared with the CT treatment as it has also been
480 observed in other studies in the Mediterranean area (Álvaro-Fuentes et al., 2009; Plaza-
481 Bonilla 2010, 2013b). However, the greater macroaggregate water-stability under NT
482 was not followed by a lower production of N₂O as suggested by Wrage et al. (2001).
483 Extrapolating our results to a structured soil, the greater macroaggregate water-stability
484 found under NT could imply a more interconnected porous space in the soil matrix. This
485 could lead to a greater aeration and reduced N₂O emissions in NT when compared with
486 CT.

487 Macroaggregate CO₂ emissions were influenced by fertilization type: when the
488 incubation was finished (i.e., after 840 hours), the macroaggregate CO₂ losses under the
489 organic fertilization treatment were higher than under the mineral fertilization treatment.
490 Moreover, the interaction between tillage and N fertilization types also affected the CO₂
491 produced by macroaggregates. Thus, under CT the CO₂ emissions were greater with the
492 use of organic fertilizer compared with either the use of mineral fertilizers or in absence
493 of fertilization. The application of pig slurry usually enhances the amount of readily
494 decomposable C compounds in the soil (Arcara et al., 1999; Sánchez-Martín et al.,
495 2008; Yang et al., 2003). This fact was observed in our experiment, in which a higher
496 WSC content was measured in the macroaggregates of the pig slurry treatment. A
497 similar trend was observed in macroaggregate water-stability. In this case, unlike the CT
498 treatment, the NT treatment did not show an interaction with the type of fertilizer on
499 macroaggregate water-stability. Contrarily, the application of organic fertilizer under
500 CT led to greater proportion of water-stable macroaggregates than the control treatment.
501 These findings suggest that the use of NT buffers the effects of the application of
502 organic fertilizers on the increase of macroaggregate stability (Plaza-Bonilla et al.,
503 2013a).

504 We found significant linear relationships between the cumulative CO₂ production and
505 (i) the macroaggregate-C concentration (R^2 : 0.21; P : 0.016) and (ii) the proportion of
506 water-stable macroaggregates (R^2 : 0.19; P : 0.021). However, when both variables (i.e.,
507 macroaggregate-C concentration and proportion of water-stable macroaggregates) were
508 included in a stepwise procedure in order to analyze their relationship with CO₂
509 production, no statistical significance was found. This finding suggests that the
510 relationship found between CO₂ production and macroaggregate stability was due to a

511 greater C concentration in those macroaggregates that are more water-stable resulting in
512 a greater production of CO₂.

513 In contrast to tillage, the different N fertilization treatments had no significant effects on
514 CH₄ emission. However, a trend (not significant) to CH₄ uptake could be observed in
515 Figure 2a in the control treatment and near zero emissions in the organic treatment.
516 Ammonium has been reported to be a competitive inhibitor of CH₄ oxidation
517 (Whittenbury et al., 1970). Interestingly, the uptake of CH₄ that we observed in the NT
518 and control treatments began after the first 72 hours of incubation and coincided with
519 the reduction in the rate of N₂O emissions. Hütsch (1998b) pointed out that CH₄
520 metabolism only begins when the nitrification process is almost completed. That
521 conclusion would explain the time-lapse that we found until the CH₄ uptake began in
522 the NT and control treatments.

523 In our experiment, fertilization type did not lead to differences in the N₂O produced by
524 soil macroaggregates. However, a trend to lower emissions under the control treatment
525 and higher emissions under the organic fertilization with pig slurry was observed (Fig.
526 2c). It is already known that the denitrification process is intensified under the presence
527 of easily decomposable C fractions such as WSC (Arcara et al., 1999). Thus, the
528 application of organic wastes, such as animal manure, usually enhances N₂O emissions
529 when compared with inorganic fertilizers (Heller et al., 2010) due to their easily
530 decomposable C content and sufficient mineral N to activate the population of
531 denitrifiers in soil (Johnson et al., 2007; Sánchez-Martín et al., 2008). Our results show
532 the relationship between the initial ammonium concentration in soil macroaggregates
533 and their N₂O production. That relationship could be explained by the role played by the
534 NH₄⁺ ion in the nitrification and denitrification processes. Ammonium oxidation is the

535 first step in the nitrification process that produces NO_3^- , which in turn is the most
536 important ion involved in the denitrification process.

537 *N₂O production by soil macroaggregates as affected by the type of N fertilization*

538 At 15% soil moisture, the nitrification process was the predominant N₂O producer,
539 while at 30% soil moisture the denitrification process emitted nine times more N (as the
540 sum of N₂O and N₂) than nitrification. These results agree with the conceptual model
541 developed by Bouwman (1998) about N₂O emissions fractionation from nitrification
542 and denitrification processes as a function of water-filled pore space. Although in small
543 amounts, the denitrification process lead to N₂O production in the macroaggregates
544 incubated at 15% soil moisture content. This could be related with the presence of
545 anaerobic microsites within the macroaggregates. Sexstone et al. (1985) quantified
546 oxygen profiles in wet aggregates and found anaerobic centers in all the aggregates that
547 denitrified. However, aerobic denitrification could have also occurred. As in other
548 studies (Bandibas et al., 1994; Diba et al., 2011; Liu et al., 2007), we observed an
549 important increase of the N₂O evolved when doubling soil moisture. This finding
550 demonstrates the role played by the absence of oxygen as electron acceptor on
551 nitrification and denitrification processes (Bouwman, 1998).

552 The combination of higher NO_3^- concentration and greater WSC in the organic N
553 treatment would explain the greater amount of N₂O evolved due to the denitrification
554 process when compared to the control treatment (Burford and Bremner, 1975;
555 Mulvaney et al., 1997). Contrarily, the greater N₂O loss from nitrification in the mineral
556 N treatment cannot be explained by a higher NH_4^+ concentration before the incubation,
557 a fact that could be related to a greater organic N mineralization during the incubation
558 that could have increased the amount of mineral N susceptible of being nitrified.

559 Different authors have observed a greater mineralization in N fertilized soils compared
560 with soils without N fertilization (Hatch et al., 2000; Zhang et al., 2012). Another
561 hypothesis could be a more efficient nitrification process in the control treatment in
562 comparison with the mineral treatment that would explain the lower N₂O emissions
563 found in the control. Nemergut et al. (2008) and Ramirez et al. (2012) found changes in
564 soil microbial community when using repeated application of mineral fertilizer when
565 compared to unfertilized soils. Thus, it could be hypothesized that the mineral fertilizer
566 applications in our field experiment could have led to changes in the microbial
567 community structure with higher nitrification efficiency in the unfertilized treatment.
568 Furthermore, our data shows a similar N₂O emission in the mineral and the organic
569 fertilization treatments. This suggests that readily decomposable C was not a limiting
570 factor to denitrification in the mineral treatment. The lack of differences between
571 fertilization treatments on macroaggregate N₂O production after the first 24 hours of
572 incubation in Experiment 1 corroborates this hypothesis.

573 **Conclusions**

574 Tillage and N fertilization treatments affected the production of GHG at the soil
575 macroaggregate scale due to changes in C and N substrates within macroaggregates.
576 Moreover, the different methanogenic and methanotrophic activities found in the tillage
577 treatments suggest changes in porosity and anaerobic conditions within soil
578 macroaggregates when either conventional tillage or no-tillage are used. Easily
579 decomposable C compounds associated with the organic fertilization together with the
580 presence of nitrate stimulated the denitrifying activity. The use of mineral and organic
581 fertilizers leads to differences in the relative importance of the nitrification and the
582 denitrification processes in the production of N₂O by soil macroaggregates: while N₂O
583 losses due to the nitrification process were preponderant in the mineral fertilization
584 treatment, denitrification N₂O losses had a higher importance under organic fertilization
585 due to a higher presence of C-labile compounds. A significant effect of the interaction
586 between tillage and N fertilization treatments on CO₂ production, with higher emissions
587 under CT when applying organic fertilizers and no differences between types of
588 fertilizers on CO₂ emissions under NT, demonstrated the capacity of NT aggregates to
589 protect C. Our study shows that tillage and N fertilization and their interaction play a
590 major role in GHG production from soil macroaggregates due to their impact on the soil
591 mineral and organic substrates that regulate the microbial activity.

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795 **Figure captions**

796 **Fig. 1** Cumulative CH₄ (A), CO₂ (B) and N₂O (C) production from dry-sieved
797 macroaggregates (0.250-8 mm) as affected by tillage (CT, conventional tillage; NT, no-
798 tillage). * For each sampling time, values are significantly different at P<0.05.

799 **Fig. 2** Cumulative CH₄ (A), CO₂ (B) and N₂O (C) production from dry-sieved
800 macroaggregates (0.250-8 mm) as affected by N fertilization (0, control; mineral N at
801 150 kg N ha⁻¹ and organic N with pig slurry at 150 kg N ha⁻¹). For each sampling time
802 different letters indicate significant differences between N fertilization treatments at
803 P<0.05.

804 **Fig. 3** Regression analysis between the initial concentration of NH₄⁺-N and the
805 cumulative N₂O-N emissions after 48 hours of incubation of dry-sieved
806 macroaggregates (0.250-8 mm).

807 **Fig. 4** Proportion of sand-free water-stable macroaggregates (0.250-8 mm) as affected
808 by tillage (CT, conventional tillage; NT, no-tillage) and N fertilization treatments
809 (control without fertilization; mineral at 150 kg N ha⁻¹; organic at 150 kg N ha⁻¹) at the
810 end of the incubation period (after 840 hours). Different letters indicate significant
811 differences between tillage and fertilization treatments at P<0.05. * Indicate significant
812 differences between tillage treatments at P<0.05.

813 **Table 1** General characteristics of the experimental site. Soil properties were measured
 814 in the Ap horizon (0-30 cm depth) at the beginning of the experiment.

Elevation (masl)	395
Mean air temperature (°C)	13.4
Annual precipitation (mm)	327
Annual ETo (mm)	1197
Soil classification [†]	Typic calcixerept
pH (H ₂ O, 1:2.5)	8.0
Organic C (g kg ⁻¹)	15.6
Organic N (g kg ⁻¹)	1.4
EC 1:5 (dS m ⁻¹)	1.0
CaCO ₃ eq. (%)	28.9
Particle size distribution (%)	
Sand (2000-50 µm)	6.2
Silt (50-2 µm)	63.3
Clay (<2 µm)	30.5

815

816 [†]According to the USDA classification (Soil Survey Staff, 1994).

817 **Table 2** Analysis of variance of the fluxes of CH₄, CO₂ and N₂O from dry-sieved
818 macroaggregates (μg CH₄-C, CO₂-C and N₂O-N kg⁻¹ macroaggregates h⁻¹, respectively)
819 as affected by tillage (CT, conventional tillage; NT, no-tillage), N fertilization
820 treatments (0, control; mineral N at 150 kg N ha⁻¹ and organic N with pig slurry at 150
821 kg N ha⁻¹), sampling time, and their interactions. Values are the means of all samplings
822 (0, 4, 12, 24, 48, 72, 192, 384, 504, 672 and 840 hours after the beginning of the
823 incubation).

	CH ₄ fluxes	CO ₂ fluxes	N ₂ O fluxes
Tillage (T)	*	n.s.	n.s.
CT	0.073 a¶	1333.67	0.754
NT	-0.207 b	1406.19	0.919
N fertilization (N)	n.s.	*	n.s.
Control	-0.175	1217.45 b	0.556
Mineral	0.074	1199.36 b	0.866
Organic	-0.096	1669.39 a	1.042
Sampling time (t)	**	***	***
T x N	n.s.	*	n.s.
CT-Control	0.046	1021.40 d	0.458
CT-Mineral	0.214	1155.13 dc	0.676
CT-Organic	-0.040	1824.48 a	1.127
NT-Control	-0.470	1478.84 ab	0.686
NT-Mineral	-0.066	1243.58 bcd	1.056
NT-Organic	-0.153	1514.30 abc	0.956
T x t	***	n.s.	n.s.
N x t	n.s.	*	n.s.
N x T x t	n.s.	n.s.	n.s.

824

825 n.s.: not significant; **P*<0.05; ***P*<0.01; *** *P*<0.001

826 ¶ For each gas and treatment, different letters indicate significant differences between
827 treatments at *P*<0.05.

828 **Table 3** Organic C concentration (OC, g kg⁻¹) and mineral N content (nitrate, NO₃⁻, and ammonium, NH₄⁺, in mg kg⁻¹) of dry-sieved
829 macroaggregates (0.250-8 mm) before (0 hours) and after (840 hours) incubation, microbial biomass C and N (MBC and MBN, respectively; mg
830 C or N kg⁻¹) and water-soluble C (WSC; mg C kg⁻¹) after the incubation (840 hours), and % of variation of C, nitrate and ammonium during the
831 incubation as affected by tillage (CT, conventional tillage; NT, no-tillage) and N fertilization treatments (0, control; mineral N at 150 kg N ha⁻¹;
832 and organic N with pig slurry at 150 kg N ha⁻¹), and their interaction.

Treatment	0 hours			840 hours						% variation 840-0 hours		
	OC	NO ₃ ⁻	NH ₄ ⁺	OC	NO ₃ ⁻	NH ₄ ⁺	MBC	MBN	WSC	OC	NO ₃ ⁻	NH ₄ ⁺
CT	6.07 (0.8)	110.4 (83.1) a¶	20.1 (15.1)	5.78 (0.8)	147.9 (72.7)	1.7 (0.7)	865.8 (232.4)	228.2 (115.6)	197.2 (43.1) b	-6.41 (3.7)	59.0 (68.9) b	-79.0 (28.9) b
NT	6.16 (0.9)	92.6 (56.0) b	13.3 (12.4)	5.72 (0.8)	141.9 (58.3)	2.0 (0.4)	954.4 (236.9)	236.5 (142.5)	234.9 (71.9) a	-7.78 (5.6)	100.8 (92.5) a	-62.3 (35.9) a
Control	5.54 (0.4) b	24.1 (4.5) c	3.0 (0.7) c	5.21 (0.6) b	87.3 (56.9) b	2.0 (0.8)	893.0 (278.4)	177.0 (70.98) b	199.5 (16.7) b	-6.07 (5.1)	182.2 (58.8) a	-32.5 (30.6) a
Mineral	5.94 (0.9) b	180.0 (43.4) a	19.0 (11.5) b	5.46 (0.5) b	183.6 (66.3) a	1.8 (0.5)	977.9 (213.1)	182.6 (96.5) b	174.3 (33.3) b	-7.61 (5.3)	7.1 (37.9) c	-86.8 (10.4) b
Organic	6.87 (0.5) a	100.4 (16.4) b	28.1 (11.7) a	6.60 (0.6) a	163.6 (12.3) a	1.7 (0.4)	859.4 (219.9)	337.5 (140.1) a	274.3 (68.5) a	-8.05 (3.8)	65.9 (22.9) b	-92.7 (4.4) b
CT-Control	5.51 (0.2)	26.0 (5.5) d	3.4 (0.8)	5.19 (0.2)	104.3 (82.2)	1.9 (1.3)	688.3 (81.3)	123.2 (44.7)	191.7 (19.0)	-5.87 (3.3)	132.2 (30.2) b	-49.2 (35.7)
CT-Mineral	5.62 (0.3)	214.4 (28.7) a	19.5 (4.2)	5.29 (0.1)	179.5 (92.4)	1.3 (0.2)	1023.2 (272.6)	208.9 (50.9)	156.5 (18.4)	-5.61 (4.7)	-13.8 (45.8) e	-93.0 (1.1)
CT-Organic	7.08 (0.4)	90.9 (8.3) c	37.4 (6.4)	6.86 (0.5)	159.9 (6.1)	1.9 (0.2)	885.9 (203.6)	352.6 (92.6)	243.4 (31.7)	-9.08 (2.2)	77.0 (16.6) c	-94.8 (0.8)
NT-Control	5.56 (0.6)	22.2 (2.6) d	2.5 (0.3)	5.23 (0.8)	70.4 (5.8)	2.1 (0.1)	1097.8 (249.9)	230.8 (45.1)	207.3 (11.3)	-6.27 (7.0)	219.7 (43.9) a	-15.7 (13.0)
NT-Mineral	6.27 (1.2)	145.7 (20.7) b	18.6 (17.0)	5.62 (0.8)	187.8 (40.7)	2.3 (0.1)	932.5 (161.6)	156.2 (131.5)	192.1 (37.5)	-9.61 (5.6)	28.1 (10.1) de	-80.5 (12.1)
NT-Organic	6.66 (0.6)	109.9 (17.8) c	18.7 (6.5)	6.34 (0.6)	167.4 (16.7)	1.6 (0.5)	832.9 (263.6)	322.5 (191.4)	305.2 (86.0)	-7.36 (5.0)	54.7 (24.8) cd	-90.5 (5.6)

833 ¶ For each variable, different letters indicate significant differences between treatments at P<0.05. Values between parentheses are the standard deviations of
834 the mean.

835 **Table 4** R^2 coefficients of the linear relationships between carbon dioxide (CO₂) and nitrous oxide (N₂O) production in conventional tillage (CT)
 836 and no-tillage (NT) macroaggregates at different times of the incubation period.

Tillage system	Sampling (hours)										
	0	4	12	24	48	72	192	384	504	672	840
CT	n.s.	0.71***	0.56**	0.61**	0.76***	0.33*	0.40*	n.s.	n.s.	n.s.	n.s.
NT	n.s.	0.72***	n.s.	n.s.	n.s.	n.s.	0.62**	n.s.	n.s.	n.s.	0.78***

837

838 n.s.: not significant; * $P < 0.05$; ** $P < 0.01$; *** $P < 0.001$

839 **Table 5** Mineral N content (nitrate, NO_3^- , and ammonium, NH_4^+ , in mg kg^{-1}) and water-
 840 soluble C (WSC; mg C kg^{-1}) of dry-sieved macroaggregates (0.250-8 mm) before
 841 incubation, as affected by N fertilization treatments (0, control; mineral N at 150 kg N
 842 ha^{-1} , and organic N with pig slurry at 150 kg N ha^{-1}).

Treatments	NH_4^+	NO_3^-	WSC
Control	1.88	20.19 b¶	90.08 b
Mineral	2.13	85.22 a	92.45 b
Organic	2.58	88.96 a	114.47 a

843

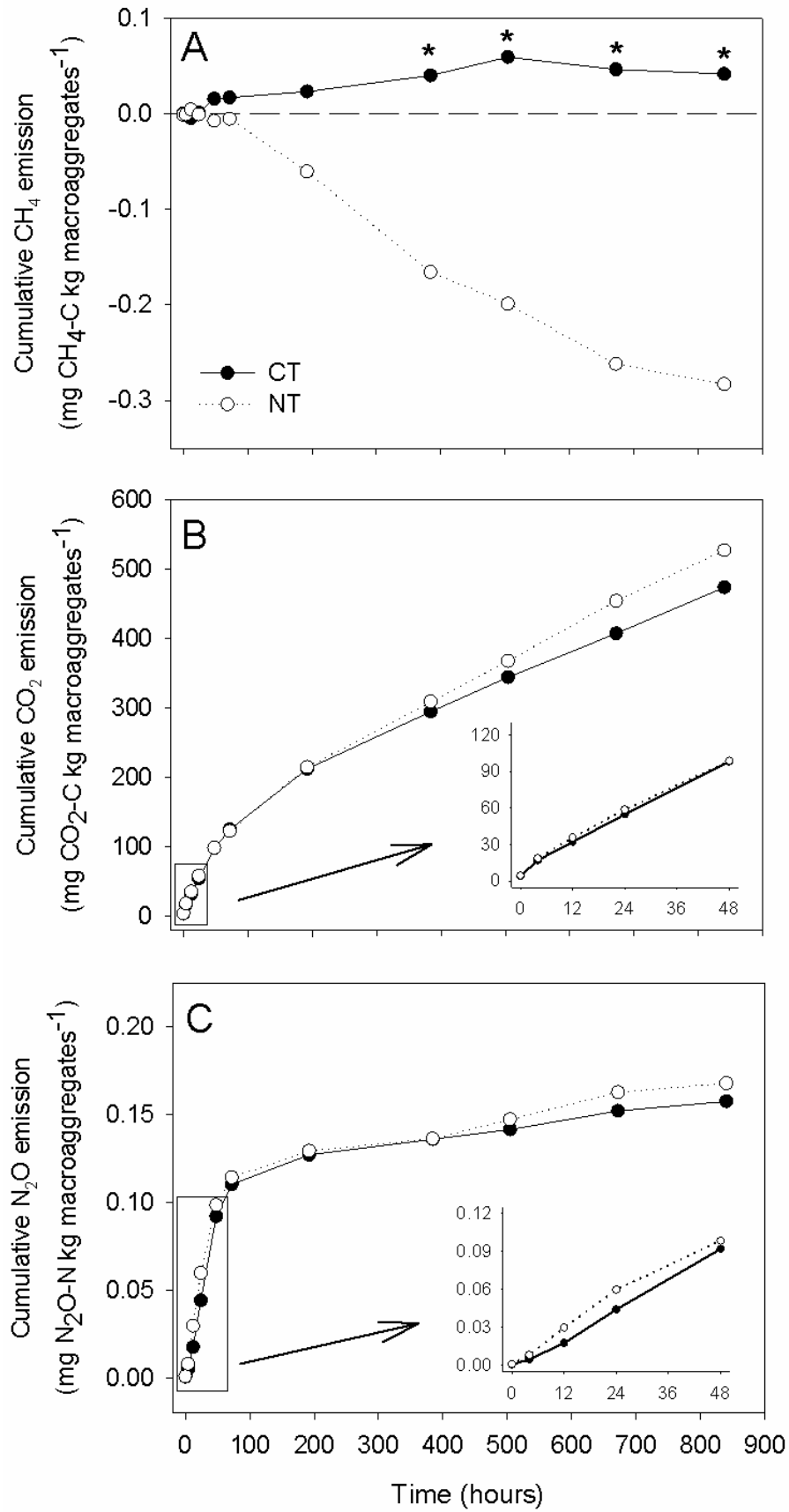
844 ¶ For each variable, different letters indicate significant differences between N fertilization
 845 treatments at $P < 0.05$.

846 **Table 6** Analysis of variance of the production of N₂O (mg N₂O-N kg⁻¹
 847 macroaggregates h⁻¹) from nitrification and denitrification processes and N₂ from
 848 denitrification process of dry-sieved macroaggregates as affected by N fertilization
 849 treatments (0, control; mineral N at 150 kg N ha⁻¹, and organic N with pig slurry at 150
 850 kg N ha⁻¹), soil moisture (15 and 30% gravimetric water content), and their interactions.

Effects	Nitrification-N ₂ O	Denitrification-N ₂ O	Denitrification-N ₂
N fertilization (N)	*	*	n.s.
Control	122.51 b¶	180.06b	1949.91
Mineral	455.25 a	303.12 ab	1436.76
Organic	247.90 ab	381.07 a	1752.38
Soil moisture (SM)	***	***	***
15%	115.06 b	69.68 b	5.32 b
30%	500.84 a	506.49 a	4120.01 a
NxSM	n.s.	n.s.	n.s.
Control – 15%	87.15	56.94	29.33
Control – 30%	193.22	303.18	4254.60
Mineral – 15%	159.37	73.39	21.0
Mineral – 30%	751.14	532.85	4268.27
Organic – 15%	60.13	78.70	0
Organic – 30 %	341.78	683.44	3896.46

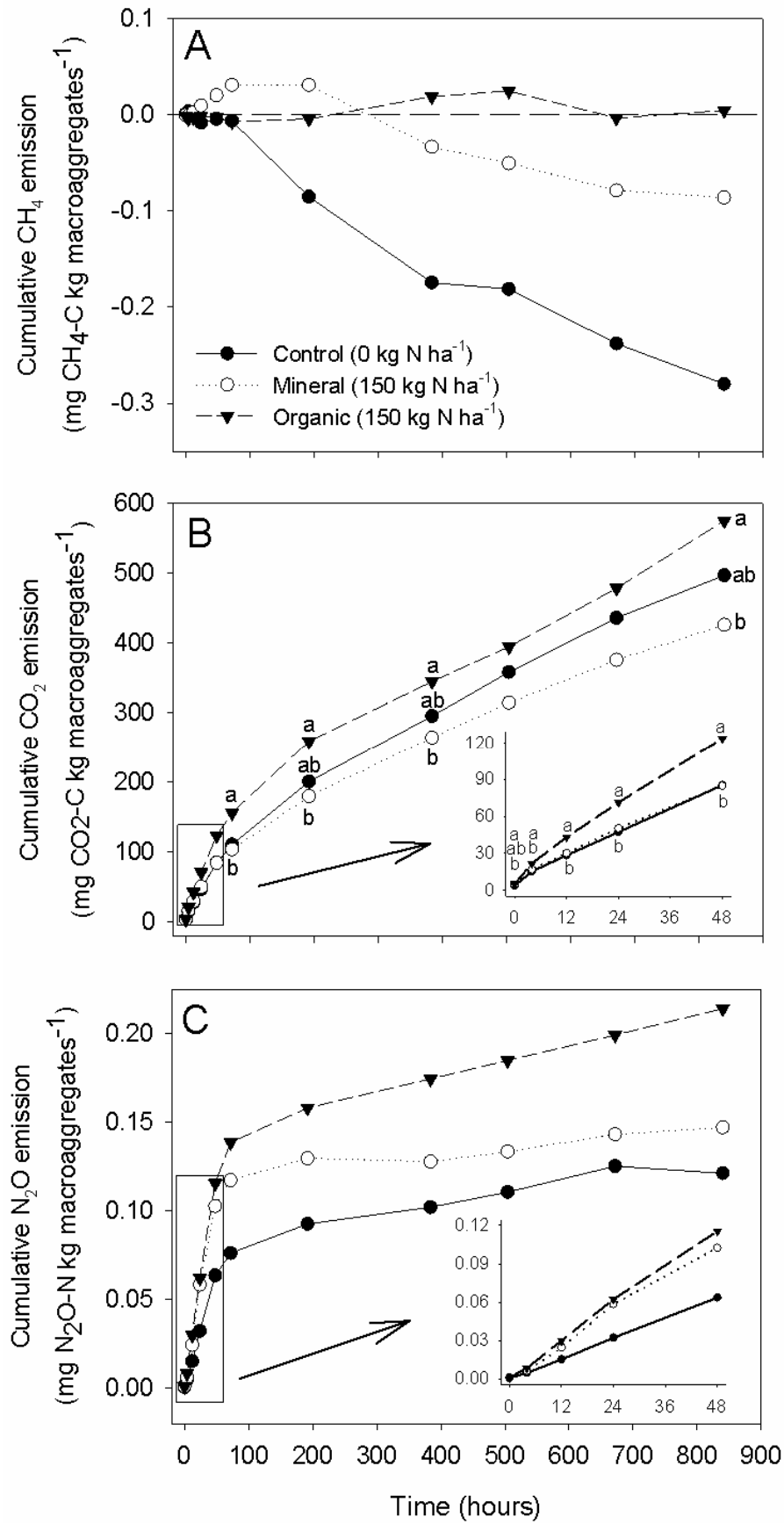
851 n.s.: not significant; **P*<0.1; ***P*<0.01; *** *P*<0.001

852 ¶ For each process, different letters indicate significant differences between N fertilization or
 853 moisture treatments at *P*<0.1.



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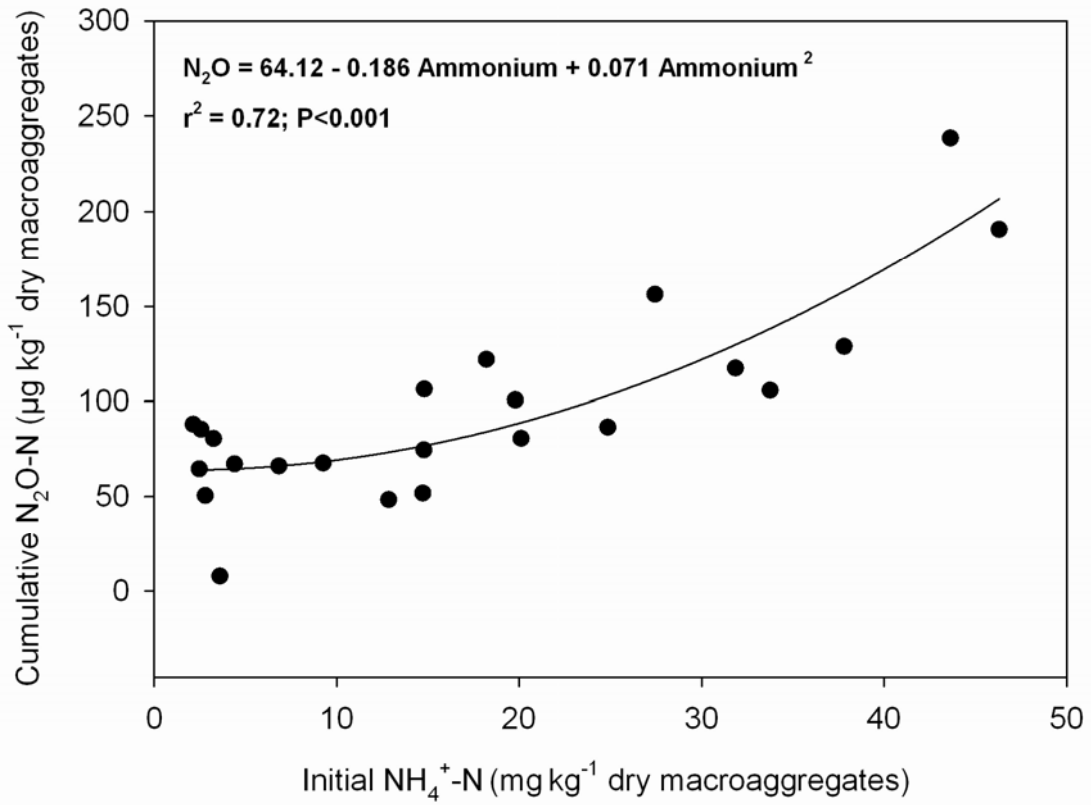
855 **Fig 1.**



856

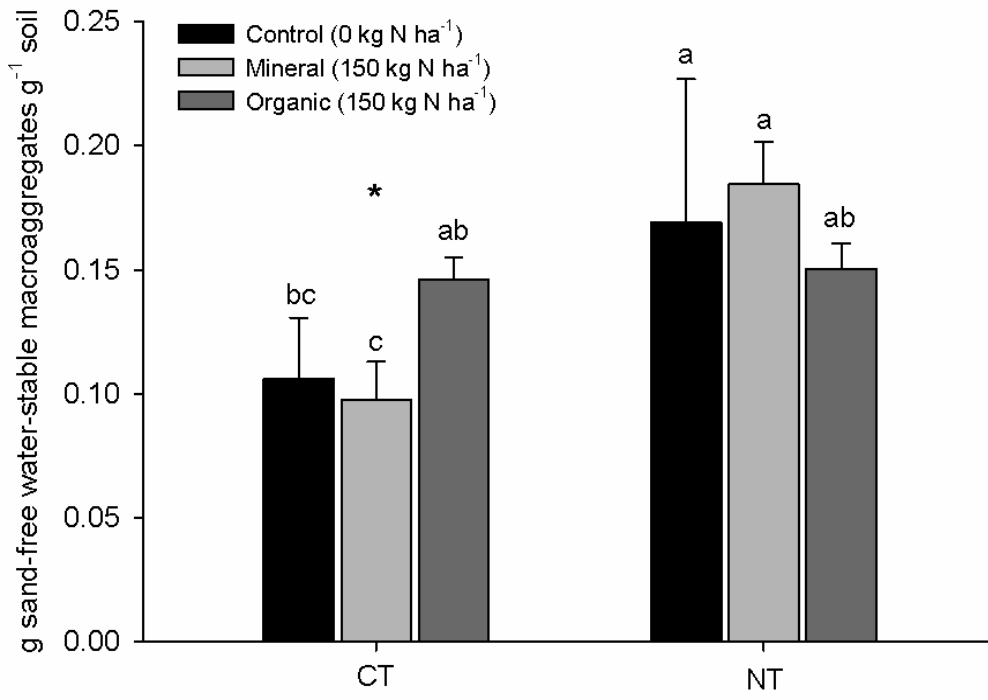
857 **Fig. 2**

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859

860 **Fig. 3**



862

863 **Fig. 4**